

UNIT 10: PERIODIC TABLE **CHAPTER 10: PERIODIC TABLE** **PERIODIC TABLE**

Section 10.1 (continued)

In your notebook, read aloud the modern periodic table.

Copy the letterheads in the boxes on the left below from the periodic table to complete the table on the right.

atomic mass	_____
atomic number	_____
electron configuration	_____
chemical name	_____
chemical symbol	_____

Put each item in Column A, under the letter of the matching item in Column B.

Column A

Column B

- | | |
|---|----------------------------------|
| _____ 18. A column in the periodic table | _____ a. metals |
| _____ 19. A row in the periodic table | _____ b. group |
| _____ 20. Group 1 elements | _____ c. period |
| _____ 21. Elements that are shiny and conduct electricity | _____ d. representative elements |
| _____ 22. Group 8 elements | _____ e. transition elements |

In the space on the left, write each of the statements in terms of the statement in bold. Change the underlined word or phrase to make it true.

- _____ 23. There are two representative elements of platinum.
- _____ 24. Most nonmetals form diatomic molecules in the periodic table (or naturally).
- _____ 25. Group 1A elements except for hydrogen are located on the left side.
- _____ 26. Group 2A elements are the alkaline earth metals.
- _____ 27. Group 7A elements are highly reactive nonmetals known as halogens.
- _____ 28. Group 8A elements are (or) unreactive elements known as noble gases.
- _____ 29. Alkaloids have properties of both metals and more reactive metals.

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**Margaret L. Lial, John Hornsby, Terry
McGinnis, Abby Tanenbaum**

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