

Chemical Equations and Reactions

SECTION 2

SHORT ANSWER Answer the following questions in the space provided.

1. Match the equation type on the left to its representation on the right.

- | | |
|----------------------------------|-----------------------------------|
| <u> c </u> synthesis | (a) $AX + BY \rightarrow AY + BX$ |
| <u> d </u> decomposition | (b) $A + BX \rightarrow AX + B$ |
| <u> b </u> single-displacement | (c) $A + B \rightarrow AX$ |
| <u> a </u> double-displacement | (d) $AX \rightarrow A + X$ |

2. c In the reaction described by the equation $2Al(s) + 3Fe(NO_3)_2(aq) \rightarrow 3Fe(s) + 2Al(NO_3)_3(aq)$, iron has been replaced by
 (a) nitrate. (c) **aluminum.**
 (b) water. (d) nitrogen.

3. a Of the following chemical equations, the only reaction that is both synthesis and combustion is
 (a) **$C(s) + O_2(g) \rightarrow CO_2(g)$.**
 (b) $2C_4H_{10}(l) + 13O_2(g) \rightarrow 8CO_2(g) + 10H_2O(l)$.
 (c) $6CO_2(g) + 6H_2O(g) \rightarrow C_6H_{12}O_6(aq) + 6O_2(g)$.
 (d) $C_6H_{12}O_6(aq) + 6O_2(g) \rightarrow 6CO_2(aq) + 6H_2O(l)$.

4. b Of the following chemical equations, the only reaction that is both combustion and decomposition is
 (a) $S(s) + O_2(g) \rightarrow SO_2(g)$.
 (b) **$2C_4H_{10}(l) + 13O_2(g) \rightarrow 8CO_2(g) + 10H_2O(l)$.**
 (c) $2H_2O_2(l) \rightarrow 2H_2O(l) + O_2(g)$.
 (d)

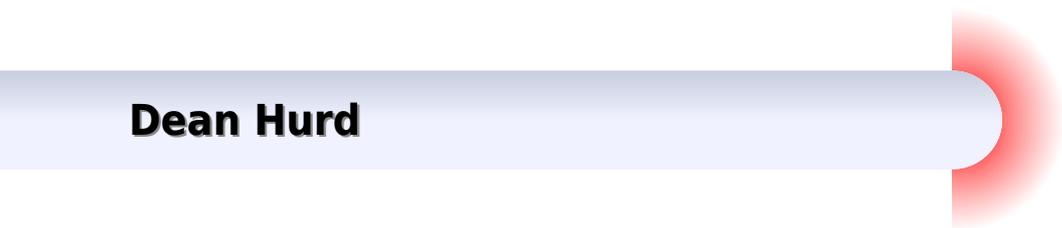
5. Identify the products when the following substances decompose:

- Its separate elements** _____ a binary compound
- Metal oxide + water** _____ most metal hydroxides
- Metal oxide + carbon dioxide** _____ a metal carbonate
- Water + sulfur dioxide** _____ the acid H_2SO_3

6. The complete combustion of a hydrocarbon in excess oxygen yields the products

Chapter Review Chemical Equations And Reactions Answers

Dean Hurd



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Table of Contents Chapter Review Chemical Equations And Reactions Answers

1. Understanding the eBook Chapter Review Chemical Equations And Reactions Answers
 - The Rise of Digital Reading Chapter Review Chemical Equations And Reactions Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chapter Review Chemical Equations And Reactions Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chapter Review Chemical Equations And Reactions Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter Review Chemical Equations And Reactions Answers
 - Personalized Recommendations
 - Chapter Review Chemical Equations And Reactions Answers User Reviews and Ratings
 - Chapter Review Chemical Equations And Reactions Answers and Bestseller Lists

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 - Chapter Review Chemical Equations And Reactions Answers eBook Subscription Services
 - Chapter Review Chemical Equations And Reactions Answers Budget-Friendly Options
6. Navigating Chapter Review Chemical Equations And Reactions Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Chapter Review Chemical Equations And Reactions Answers Compatibility with Devices
 - Chapter Review Chemical Equations And Reactions Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chapter Review Chemical Equations And Reactions Answers
 - Highlighting and Note-Taking Chapter Review Chemical Equations And Reactions Answers
 - Interactive Elements Chapter Review Chemical Equations And Reactions Answers
8. Staying Engaged with Chapter Review Chemical Equations And Reactions Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chapter Review Chemical Equations And Reactions Answers
9. Balancing eBooks and Physical Books Chapter Review Chemical Equations And Reactions Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chapter Review Chemical Equations And Reactions Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chapter Review Chemical Equations And Reactions Answers
 - Setting Reading Goals Chapter Review Chemical Equations And Reactions Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chapter Review Chemical Equations And Reactions Answers
 - Fact-Checking eBook Content of Chapter Review Chemical Equations And Reactions Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
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 - Interactive and Gamified eBooks

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