

## Double Replacement Reactions and Solubility

**Post-Lab Questions:** Review your Observations Table. Complete AND balance each equation below. If your results yielded a precipitate, then write the net ionic equation in the box to the right. MAKE SURE TO SHOW STATES/PHASES for every reactant and product! Remember that all nitrates are soluble in water. If you did not see signs of a precipitate, then write NONE in the box to the right.

1)	$\text{Na}_2\text{CO}_3(\text{aq}) + \text{Al}(\text{NO}_3)_3(\text{aq}) \rightarrow 6\text{NaNO}_3(\text{aq}) + \text{Al}(\text{OH})_3(\text{s})$	None
2)	$\text{NaCl}(\text{aq}) + \text{NH}_4\text{NO}_3(\text{aq}) \rightarrow \text{NaNO}_3(\text{aq}) + \text{NH}_4\text{Cl}(\text{aq})$	None
3)	$\text{NaI}(\text{aq}) + \text{AgNO}_3(\text{aq}) \rightarrow \text{NaNO}_3(\text{aq}) + \text{AgI}(\text{s})$	$\text{Ag}^+(\text{aq}) + \text{I}^-(\text{aq}) \rightarrow \text{AgI}(\text{s})$
4)	$\text{Na}_3\text{PO}_4(\text{aq}) + \text{Cu}(\text{NO}_3)_2(\text{aq}) \rightarrow 6\text{NaNO}_3(\text{aq}) + (\text{Cu}_3(\text{PO}_4)_2(\text{s}))$	$3\text{Cu}^{2+}(\text{aq}) + 2\text{PO}_4^{3-}(\text{aq}) \rightarrow \text{Cu}_3(\text{PO}_4)_2(\text{s})$

Use your observations table to answer the following. **ALL ions must include a symbol AND a charge or it is wrong!**

5 a) The symbols and charges of the anions that produced the fewest (<3) precipitates are \_\_\_\_\_

b) Salts of these anions may be described as SOLUBLE except for those with the cations \_\_\_\_\_

6a) The symbols and charges of the anion that produced the most (>5) precipitates are \_\_\_\_\_

b) Salts of these anions may be described as INSOLUBLE except for those with the cations \_\_\_\_\_

7. Although you did not test the bromide ion ( $\text{Br}^-$ ), would you expect most bromide salts ( $\text{AlBr}_3$ ,  $\text{BaBr}_2$ ,  $\text{CaBr}_2$ , etc) to be soluble or insoluble? Hint: Locate bromine on the periodic table. \_\_\_\_\_

8. Explain your answer in #7 \_\_\_\_\_

9. What cation combined with bromide is an exception to the rule in #7? \_\_\_\_\_

10-14 Summarize the general solubility rules of ionic compounds in the table below. Two have been done for you.

Ion	Generally soluble or insoluble?	Exceptions
9) carbonates		
10) chlorides, bromides, iodides		
11) hydroxides		
nitrates	Soluble	None
12) phosphates		

# Double Replacement Reactions And Solubility Answer Key

**B Lingard**



## **Double Replacement Reactions And Solubility Answer Key:**

**General Chemistry** Donald A. McQuarrie, Stanley Gill, 2011-06-15 This Fourth Edition of McQuarrie's classic text offers a thorough revision and a quantum leap forward from the previous edition Taking an atoms first approach it promises to be another ground breaking text in the tradition of McQuarrie's many previous works This outstanding new text available in a soft cover edition offers professors a fresh choice and outstanding value *Regents Exams and Answers: Chemistry--Physical Setting Revised Edition* Barron's Educational Series, Albert Tarendash, 2021-01-05 Barron's Regents Exams and Answers Chemistry provides essential practice for students taking the Chemistry Regents including actual recently administered exams and thorough answer explanations for all questions This book features Eight actual administered Regents Chemistry exams so students can get familiar with the test Thorough explanations for all answers Self analysis charts to help identify strengths and weaknesses Test taking techniques and strategies A detailed outline of all major topics tested on this exam A glossary of important terms to know for test day **The Science Teacher's Activity-A-Day, Grades 5-10** Pam Walker, Elaine Wood, 2010-10-05 A hands on and fun filled resource for teaching science to middle and high school students New in the 5 Minute Fundamentals Series The Science Teacher's Activity A Day Grades 6 12 includes 180 easy five minute hook or sponge activities to capture learners attention and introduce lessons Divided into three units Physical Science Life Science and Earth and Space Science the activities cover topics based on the National Science Education Standards All the book's activities can be done with materials that are inexpensive and easy to find Includes quick and fun sponge activities that are designed to engage students All the activities take about 5 minutes to complete The Science Teacher's Activity a Day is an ideal resource for middle and high school science teachers **General, Organic, and Biochemistry** Katherine J. Denniston, Joseph J. Topping, Robert L. Caret, 2001 [AP Chemistry Premium, 2026: Prep Book with 6 Practice Tests + Comprehensive Review + Online Practice](#) Barron's Educational Series, Neil D. Jespersen, Pamela Kerrigan, 2025-07 Be prepared for exam day with Barron's Trusted content from AP experts Barron's AP Chemistry Premium 2026 includes in depth content review and practice It's the only book you'll need to be prepared for exam day Written by Experienced Educators Learn from Barron's all content is written and reviewed by AP experts Build your understanding with comprehensive review tailored to the most recent changes made to the course and exam by the College Board for 2025 and beyond Get a leg up with tips strategies and study advice for exam day it's like having a trusted tutor by your side Be Confident on Exam Day Sharpen your test taking skills with 6 full length practice tests 3 in the book and 3 more online plus 3 short diagnostic tests for assessing strengths and areas for improvement and detailed answer explanations for all questions Strengthen your knowledge with in depth review covering all units on the AP Chemistry exam including the changes on removing the big ideas changing titles of units and revising topics and learning objectives Reinforce your learning with more than 300 practice questions throughout the book that cover all frequently tested topics Learn what to expect on test day with

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**Schaum's Outline of Theory and Problems of Beginning Chemistry** David Elliott Goldberg, 1999 Featuring completely updated problems and the latest terminology this study guide is the perfect aid for better grades Illustrations

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contribute to a company that cares about its community and employees 2 how would you handle a

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