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By Judy Blume



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David James Clarke



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so calculating enthalpy change of reaction in 2 $\Delta H_{\text{reaction}} = 6 \times \Delta H_{\text{f}}^{\circ}(\text{C}_2\text{H}_2) + 6 \times \Delta H_{\text{f}}^{\circ}(\text{C}_2\text{H}_4) + 6 \times \Delta H_{\text{f}}^{\circ}(\text{C}_2\text{H}_6) = 523 \text{ kJ}$

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electrons and chlorine in group 7a has 7 valence electrons so the total number of valence electrons in is the lewis structure
of is

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