

Assignment
2011-2012

CLASS IX CHEMISTRY

S.A. II

Atoms and molecules

Q1) Calculate the % of Nitrogen in urea (NH_2CONH_2).

Q2) Law of conservation of mass was discovered by

- (a) Dalton (b) Proust (c) Lavoisier (d) Richter

Q3) Two elements X & Y combine in a gaseous state to form XY in the ratio of 1: 35.5 by mass. The mass of Y which combines with 2 gram of X will be

- (a) 7.1 gram (b) 3.55 gram (c) 35.5 gram (d) 71 gram

Q4) Match the following:-

Column I	Column II
1. Sodium	a. Yellow in colour
2. Helium	b. Diatomic
3. Sulphur	c. Soft metal
4. Hydrogen	d. Noble gas

Q 5) Give one word for the following:-

- (a) Metal present in haemoglobin.
(b) Reddish brown metal used in making electrical wires.

Q 6) Write correct formulae using given ions.

anions	SO_4^{2-}	Cl^-	PO_4^{3-}	O^{2-}	CO_3^{2-}	OH^-	S^{2-}
cations							
Na^+							
Cu^{2+}							
Al^{3+}							

Q 7) What happens when : --

- (a) Solution of sodium chloride and silver nitrate are mixed together.
(b) Solution of barium chloride and sodium sulphate react together.

Q8) Calculate the molecular mass of $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$

[Cu=63.5 u, S=32 u, O=16 u, H=1 u]

Q9) The % of three elements calcium, carbon & oxygen in a sample of calcium carbonate is given as:-

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RP Manchanda



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