

## Ion Worksheet

Symbol	Ion name	Cation or Anion	More or less electrons compared to atom	Number of electrons gained or lost compared to atom
<b>Li<sup>+</sup></b>	<b>Lithium ion</b>	<b>Cation</b>	<b>Less</b>	<b>1</b>
<b>Na<sup>+</sup></b>	Sodium ion	Cation	Less	1
<b>Cl<sup>-</sup></b>	<b>Chloride ion</b>	Anion	More	1
<b>Al<sup>3+</sup></b>	<b>Aluminum ion</b>	Cation	Less	3
<b>S<sup>2-</sup></b>	Sulfide ion	Anion	More	2
<b>Mg<sup>2+</sup></b>	Magnesium ion	Cation	Less	2
<b>Fe<sup>3+</sup></b>	<b>Iron (III) ion</b>	Cation	Less	3
<b>Cu<sup>2+</sup></b>	Copper (II) ion	Cation	Less	2
<b>Cu<sup>+</sup></b>	<b>Copper (I) ion</b>	Cation	Less	1
<b>Fe<sup>2+</sup></b>	Iron (II) ion	Cation	Less	2
<b>O<sup>2-</sup></b>	<b>Oxide ion</b>	Anion	More	2
<b>N<sup>3-</sup></b>	Nitride ion	Anion	More	3
<b>Ca<sup>2+</sup></b>	<b>Calcium ion</b>	Cation	Less	2
<b>I<sup>-</sup></b>	Iodide ion	Anion	More	1
<b>K<sup>+</sup></b>	Potassium ion	Cation	Less	1
<b>Ni<sup>2+</sup></b>	<b>Nickel (II) ion</b>	Cation	Less	2

# Chapter 7 7 2 Ions Answer Key

**Steven S. Zumdahl**



## Chapter 7 7 2 Ions Answer Key:

**Carboranes** Russell N. Grimes, 2011-03-10 Carboranes Second Edition is designed as a comprehensive source of information in a field that has experienced enormous growth in both its fundamental and applied aspects in the four decades since the publication of Carboranes 1970 During this long period thousands of original research papers have appeared along with many review articles and book chapters dealing with aspects of carborane chemistry As carborane science has grown in complexity and applications have advanced steadily in areas such as medicine nanostructured and electroactive materials catalysis polymers and others the need for a monograph covering the entire area in a unified treatment has become increasingly apparent This volume has two principal objectives the first of which is to provide a readable and concise introduction to the basic principles underlying the synthesis structures reactivity and applications of carboranes and metallocarboranes at a level suitable for readers in industry and academe who are not trained in boron chemistry but find themselves working with or lecturing about carboranes Secondly the book furnishes a trove of detailed information for workers active in carborane science and associated technologies To that end it incorporates tables listing thousands of specific compounds keyed to literature references together with more than 2 000 molecular structure drawings that illuminate the accompanying discussion Thorough treatment of the synthesis structures and reactions of carboranes heterocarboranes and metallocarboranes in the first 13 chapters is followed by four chapters detailing advances in practical applications in polymer science catalysis medicine and other areas Includes over 2 000 molecular structure drawings throughout the text Features tables listing thousands of compounds with key literature references **Quantitative**

**Chemical Analysis** Daniel C. Harris, 2010-04-30 QCA is the bestselling textbook of choice for analytical chemistry It offers a modern portrait of the techniques of chemical analysis backed by a wealth of real world applications This edition features new coverage of spectroscopy and statistics new pedagogy and enhanced lecturer support **Chemistry for Changing**

**Times** John William Hill, 1984 **Colloid Chemistry** Jerome Alexander, 1929 **Outlines of Theoretical Chemistry** Frederick Hutton Getman, 1922 *Chemical Fundamentals of Geology and Environmental Geoscience* Robin Gill, 2015-01-27 Chemical principles are fundamental to the Earth sciences and geoscience students increasingly require a firm grasp of basic chemistry to succeed in their studies The enlarged third edition of this highly regarded textbook introduces the student to such geo relevant chemistry presented in the same lucid and accessible style as earlier editions but the new edition has been strengthened in its coverage of environmental geoscience and incorporates a new chapter introducing isotope geochemistry The book comprises three broad sections The first Chapters 1 4 deals with the basic physical chemistry of geological processes The second Chapters 5 8 introduces the wave mechanical view of the atom and explains the various types of chemical bonding that give Earth materials their diverse and distinctive properties The final chapters 9 11 survey the geologically relevant elements and isotopes and explain their formation and their abundances in the cosmos and the Earth

The book concludes with an extensive glossary of terms appendices cover basic maths explain basic solution chemistry and list the chemical elements and the symbols units and constants used in the book **Chemistry** George Tyler Miller,1987

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Michael J. Padilla,2002 Volumetric Analysis Dr. I.M. Kolthoff, N. Howell Furman Ph.D.,1928 The Chemistry of  
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Michael Jellum,1991 Introduction to Oceanography David A. Ross,1995 This text provides an introduction to the science  
of the world s oceans With an interdisciplinary approach it explains how the basic sciences biology chemistry geology and  
physics with some social sciences influence oceanography Ecological issues are discussed throughout *Workbook for*  
*Organic Chemistry* Jerry Jenkins,2009-12-25 With authors who are both accomplished researchers and educators Vollhardt  
and Schore s Organic Chemistry is proven effective for making contemporary organic chemistry accessible introducing  
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the substantial information presented in this course And in the sixth edition the themes of understanding reactivity  
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