

## Chapter 7: Chemical Formulas and Chemical Compounds

### I. CHEMICAL NAMES AND FORMULAS

#### A. Significance of a Chemical Formula

1. A chemical formula shows the relative number of atoms of each kind in a chemical compound, with subscripts indicating the number of atoms of one type of element.
2. In a molecular formula, the chemical formula shows the number of atoms of each element in one molecule of the compound.
  - a) For example,  $H_2O$ , the chemical formula for water, indicates the presence of 2 hydrogen atoms and 1 oxygen atom.
3. In an ionic compound, the chemical formula represents one formula unit, which is the simplest ratio of the compound, positive ions (cations) and negative ions (anions).

#### B. Monatomic Ions

1. Many main-group elements can become ions by gaining or losing electrons to have noble-gas configurations.
2. Monatomic ions are ions formed from a single atom.
  - a) Group 1 metals lose one electron; Group 2 elements lose two electrons. They will have +1 and +2 charges, respectively.
  - b) The nonmetals in Group 15, 16, and 17 gain electrons and form anions.
  - c) Metals with *d* orbitals are slightly different, as they can form +2, +3, or even +1 and +4. Some of them can even form ions of different charges.
    - (1) Refer to the chart on page 221 of Modern Chemistry published by Holt, Rinehart, and Winston.

#### C. Naming Monatomic Ions

1. Monatomic cations are identified by their normal name.
2. Monatomic anions, however, have their ending dropped and the ending , -ide is added to the root name.

#### D. Binary Ionic Compounds

1. Binary compounds are compounds composed of two elements.
2. In a binary ionic compound, the total numbers of positive charges and negative charges must be equal.

# Chapter Chemical Formulas And Compounds

**Stanley E. Manahan**



## **Chapter Chemical Formulas And Compounds:**

**Fundamentals of Chemistry** Lotta Jean Bogert, 1924 *Basics for Chemistry* David A. Ucko, 2013-09-24 Basics of Chemistry provides the tools needed in the study of General Chemistry such as problem solving skills calculation methods and the language and basic concepts of chemistry The book is designed to meet the specific needs of underprepared students Concepts are presented only as they are needed and developed from the simple to the complex The text is divided into 18 chapters each covering some particular aspect of chemistry such as matter energy and measurement the properties of atoms description of chemical bonding study of chemical change and nuclear and organic chemistry Undergraduate students will find the book as a very valuable academic material *Fundamentals of Environmental Chemistry, Third Edition* Stanley E. Manahan, 2011-03-05 Written by an expert using the same approach that made the previous two editions so successful *Fundamentals of Environmental Chemistry Third Edition* expands the scope of book to include the strongly emerging areas broadly described as sustainability science and technology including green chemistry and industrial ecology The new edition includes Increased emphasis on the applied aspects of environmental chemistry Hot topics such as global warming and biomass energy Integration of green chemistry and sustainability concepts throughout the text More and updated questions and answers including some that require Internet research Lecturers Pack on CD ROM with solutions manual PowerPoint presentations and chapter figures available upon qualifying course adoptions The book provides a basic course in chemical science including the fundamentals of organic chemistry and biochemistry The author uses real life examples from environmental chemistry green chemistry and related areas while maintaining brevity and simplicity in his explanation of concepts Building on this foundation the book covers environmental chemistry broadly defined to include sustainability aspects green chemistry industrial ecology and related areas These chapters are organized around the five environmental spheres the hydrosphere atmosphere geosphere biosphere and the anthrosphere The last two chapters discuss analytical chemistry and its relevance to environmental chemistry Manahan s clear concise and readable style makes the information accessible regardless of the readers level of chemistry knowledge He demystifies the material for those who need the basics of chemical science for their trade profession or study curriculum as well as for readers who want to have an understanding of the fundamentals of sustainable chemistry in its crucial role in maintaining a livable planet **O Level Chemistry MCQ (Multiple Choice Questions)** Arshad Iqbal, 2019-06-27 The O Level Chemistry Multiple Choice Questions MCQ Quiz with Answers PDF O Level Chemistry MCQ PDF Download Quiz Questions Chapter 1 14 Practice Tests with Answer Key IGCSE GCSE Chemistry Questions Bank MCQs Notes includes revision guide for problem solving with hundreds of solved MCQs O Level Chemistry MCQ with Answers PDF book covers basic concepts analytical and practical assessment tests O Level Chemistry MCQ PDF book helps to practice test questions from exam prep notes The O Level Chemistry MCQs with Answers PDF eBook includes revision guide with verbal quantitative and analytical past papers solved MCQs O Level Chemistry

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- Chapter 1: Introduction to Chapter Chemical Formulas And Compounds
- Chapter 2: Essential Elements of Chapter Chemical Formulas And Compounds
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- Chapter 4: Chapter Chemical Formulas And Compounds in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Chapter Chemical Formulas And Compounds. The first chapter will explore what Chapter Chemical Formulas And Compounds is, why Chapter Chemical Formulas And Compounds is vital, and how to effectively learn about Chapter Chemical Formulas And Compounds.
3. In chapter 2, this book will delve into the foundational concepts of Chapter Chemical Formulas And Compounds. This chapter will elucidate the essential principles that must be understood to grasp Chapter Chemical Formulas And Compounds in its entirety.
4. In chapter 3, this book will examine the practical applications of Chapter Chemical Formulas And Compounds in daily life. This chapter will showcase real-world examples of how Chapter Chemical Formulas And Compounds can be effectively utilized in everyday scenarios.
5. In chapter 4, this book will scrutinize the relevance of Chapter Chemical Formulas And Compounds in specific contexts. The fourth chapter will explore how Chapter Chemical Formulas And Compounds is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, this book will draw a conclusion about Chapter Chemical Formulas And Compounds. The final chapter will summarize the key points that have been discussed throughout the book.
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