

# STIOCHIOMETRY

## CHEMISTRY CHAPTER 9

### 9.1 What is Stiochiometry

A. The study of quantitative relationships between amounts of reactants used and product formed by chemical reactions.

B. Mole-Mass relationships in chemical reactions

1. Stiochiometry and balanced chemical equations.



2. Calculate Reactants and Products

$$4 \text{ mol Fe} \times \frac{55.85 \text{ g Fe}}{1 \text{ mol Fe}} = 223.4 \text{ g Fe}$$

$$3 \text{ mol O}_2 \times \frac{32.00 \text{ g O}_2}{1 \text{ mol O}_2} = 96.0 \text{ g O}_2$$

Total Mass in grams of Reactants = 319.4 g

# Chapter 9 Test Stiochiometry Key

**Amanda Stewart**



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