

## SOLUBILITY CURVE WORKSHEET KEY

Use your solubility curve graphs provided to answer the following questions.

1. What are the customary units of solubility on solubility curves? Degress Celsius and grams of solute/100g of water
2. Define solubility. A measure of how much solute can dissolve in a given amount of solvent.
3. According to the graph, the solubility of any substance changes as temperature changes.
4. List the substances whose solubility decreases as temperature increases.  $\text{NH}_3$  and  $\text{Ce}_2(\text{SO}_4)_3$ .
5. Which substance is least affected by temperature changes?  $\text{NaCl}$ .
6. How many grams of ammonium chloride ( $\text{NH}_4\text{Cl}$ ) at  $50^\circ\text{C}$ ? 50g
7.  $\text{NaCl}$  and  $\text{KClO}_3$  have the same solubility at approximately  $78^\circ\text{C}$ .
8. Which compound is least soluble in water at  $10^\circ\text{C}$ ?  $\text{KClO}_3$ .
9. How many grams of  $\text{KNO}_3$  can be dissolved at  $50^\circ\text{C}$ ? 80g
10. Are the following solutions unsaturated, saturated, or supersaturated?
  - a. 45g of  $\text{NaNO}_3$  in 100 g of water at  $30^\circ\text{C}$ . saturated
  - b. 60g of  $\text{KClO}_3$  in 100 g of water at  $90^\circ\text{C}$ . supersaturated
11. How many grams of sodium chloride,  $\text{NaCl}$  are required to saturate 100 grams of water at  $100^\circ\text{C}$ ? 40g
12. How many grams of  $\text{NaNO}_3$  are required to saturate 100 grams of water at  $75^\circ\text{C}$ ? 140g
13. How many grams of  $\text{KI}$  will saturate water at  $20^\circ\text{C}$ ? 33g
14. At what temperature would 25g of potassium chlorate ( $\text{KClO}_3$ ) dissolve?  $60^\circ\text{C}$
15. At what temperature would 60g of  $\text{NH}_4\text{Cl}$  dissolve?  $70^\circ\text{C}$
16. 89 g  $\text{NaNO}_3$  is prepared at  $30^\circ\text{C}$ .
  - a) Will all of the salt dissolve? No
  - b) What mass of  $\text{NaNO}_3$  will dissolve at this temperature? 95g
17. If 50 grams of  $\text{NH}_4\text{Cl}$  is dissolved at  $50^\circ\text{C}$ , how many additional grams  $\text{NH}_4\text{Cl}$  would be needed to make the solution saturated at  $80^\circ\text{C}$ ? 15g
18. At  $50^\circ\text{C}$ , how many grams of  $\text{KNO}_3$  will dissolve? 80g
19. At  $70^\circ\text{C}$ , how many grams of cerium (III) sulfate ( $\text{Ce}_2(\text{SO}_4)_3$ ) dissolve? 5g
20. Determine if each of the following is unsaturated, saturated, or supersaturated.
  - a. 55g of  $\text{NH}_3$  at  $20^\circ\text{C}$  supersaturated
  - b. 10g of  $\text{Ce}_2(\text{SO}_4)_3$  at  $10^\circ\text{C}$  unsaturated
  - c. 110g of  $\text{KNO}_3$  at  $60^\circ\text{C}$ . supersaturated
  - d. 65g of  $\text{NH}_4\text{Cl}$  at  $80^\circ\text{C}$ . saturated
  - e. 12g of  $\text{NH}_3$  at  $90^\circ\text{C}$ . supersaturated
  - f. 78g of  $\text{NaNO}_3$  at  $10^\circ\text{C}$ . saturated
  - g. 145g of  $\text{NaNO}_3$  at  $80^\circ\text{C}$ . saturated
  - h. 35g of  $\text{NaCl}$  at  $100^\circ\text{C}$ . unsaturated

# Answer Key To Solubility Curve 1

**American Ceramic Society**



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*A.M.A. Archives of Internal Medicine*, 1952

Dehydration of food Joint Committee of the National Association of Food Manufacturers and the American Society of Heating and Ventilation Engineers, 1923

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## **Table of Contents Answer Key To Solubility Curve 1**

1. Understanding the eBook Answer Key To Solubility Curve 1
  - The Rise of Digital Reading Answer Key To Solubility Curve 1
  - Advantages of eBooks Over Traditional Books
2. Identifying Answer Key To Solubility Curve 1
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Answer Key To Solubility Curve 1
  - User-Friendly Interface
4. Exploring eBook Recommendations from Answer Key To Solubility Curve 1
  - Personalized Recommendations
  - Answer Key To Solubility Curve 1 User Reviews and Ratings
  - Answer Key To Solubility Curve 1 and Bestseller Lists
5. Accessing Answer Key To Solubility Curve 1 Free and Paid eBooks
  - Answer Key To Solubility Curve 1 Public Domain eBooks
  - Answer Key To Solubility Curve 1 eBook Subscription Services
  - Answer Key To Solubility Curve 1 Budget-Friendly Options
6. Navigating Answer Key To Solubility Curve 1 eBook Formats

- ePub, PDF, MOBI, and More
  - Answer Key To Solubility Curve 1 Compatibility with Devices
  - Answer Key To Solubility Curve 1 Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Answer Key To Solubility Curve 1
  - Highlighting and Note-Taking Answer Key To Solubility Curve 1
  - Interactive Elements Answer Key To Solubility Curve 1
8. Staying Engaged with Answer Key To Solubility Curve 1
- Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Answer Key To Solubility Curve 1
9. Balancing eBooks and Physical Books Answer Key To Solubility Curve 1
- Benefits of a Digital Library
  - Creating a Diverse Reading Collection Answer Key To Solubility Curve 1
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Answer Key To Solubility Curve 1
- Setting Reading Goals Answer Key To Solubility Curve 1
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Answer Key To Solubility Curve 1
- Fact-Checking eBook Content of Answer Key To Solubility Curve 1
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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