

9. Which statement is TRUE about the most stable conformation of the molecule?



- (A) the *t*-butyl and chloro groups are both equatorial and the bromo group is axial
- (B) the bromo and chloro groups are both equatorial and the *t*-butyl group is axial
- (C) the *t*-butyl and bromo groups are both equatorial and the chloro group is axial
- (D) all three groups are equatorial

10. Which Newman projection depicts the lowest energy conformation of (*R*)-3-methylhexane?



11. Which are aromatic?



- (A) I and III
- (B) II and IV
- (C) II and III
- (D) I and II

12. What is the result of hydroboration of 1-methylcyclopentene, followed by oxidation?



- (A) a pair of enantiomers
- (B) a pair of diastereomers
- (C) a meso compound
- (D) two achiral compounds

13. Which molecule will have the smallest HOMO-LUMO energy gap?



14. Which act as a Bronsted base?



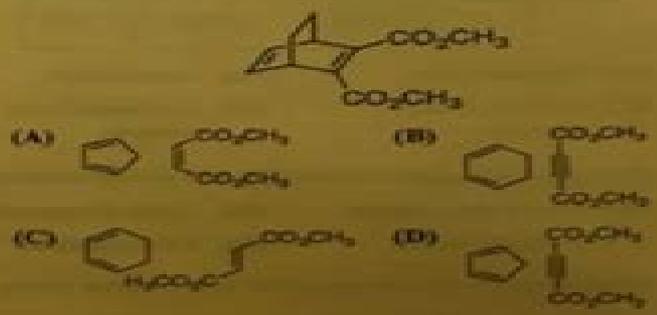
- (A) I
- (B) II
- (C) I and IV
- (D) II and III

15. What is the enantiomeric excess (*ee*) of the product mixture in this reaction?



- (A) 26%
- (B) 37%
- (C) 63%
- (D) 100%

16. Which reactants should be used to form the product?

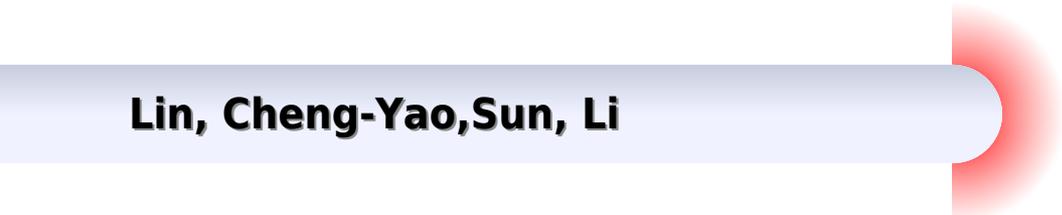


17. Which compound will be protonated to the greatest extent in  $1.0 \times 10^{-2}$  M aqueous HCl?



# Acs General Organic Biochemistry Practice Exam

**Lin, Cheng-Yao, Sun, Li**



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