

1. What are the hybridizations of atoms I and II?



- (A) I -  $sp^2$  and II -  $sp^3$   
 (B) I -  $sp^3$  and II -  $sp$   
 (C) I -  $sp^2$  and II -  $sp^2$   
 (D) I -  $sp^3$  and II -  $sp^2$

2. What is the isomeric relationship between these Newman projections?



- (A) diastereomers  
 (B) enantiomers  
 (C) constitutional isomers  
 (D) conformational isomers

3. Which is not capable of existing as a pair of enantiomers?

- (A) 1,3-dichlorobutane  
 (B) 1,2-dichloropropane  
 (C) 3-ethylpentane  
 (D) 2-bromobutane

Rank the isomeric alkenes from least to most stable.

- I pent-1-ene  
 II 2-methylbut-2-ene  
 III (*E*)-pent-2-ene  
 IV (*Z*)-pent-2-ene

- II < III < IV < I  
 < III < IV < II  
 : IV < III < II  
 - IV < III < I

5. Which alkene was used in this cleavage reaction?



- (A)
- (B)
- (C)
- (D)

6. Rank these radicals from least to most stable.



- (A) I < II < III  
 (B) II < I < III  
 (C) III < I < II  
 (D) III < II < I

7. Which compound is most soluble in water?

- (A) dipropyl ether  
 (B) 2,3-dichlorobutane  
 (C) 2,5-hexanediol  
 (D) trichloromethane

8. Which is (2S,3R)-2,3,4-trihydroxybutanal?

- (A)
- (B)
- (C)
- (D)

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**Oscar Krisen Buros**



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