



## Chapter 2: Acids, Bases and Salts

Science Worksheet

Standard: 10th

Subject: Science

Student Name: \_\_\_\_\_

Q1. Match the chemical substances given in Column (A) with their appropriate application given in Column (B):

Column (A)	Column (B)
A. Bleaching powder	1. Preparation of glass
B. Baking soda	2. Production of $H_2$ and $Cl_2$
C. Washing soda	3. Decolourisation
D. Sodium chloride	4. Antacid

1. A-2, B-1, C-4, D-3
2. A-3, B-2, C-4, D-1
3. A-3, B-4, C-1, D-2
4. A-2, B-4, C-1, D-3

Q2. 10mL of a solution of NaOH is found to be completely neutralized by 8mL of a given solution of HCl. If we take 20mL of same solution of NaOH, the amount of HCl solution required to neutralize it will be:

1. 4mL
2. 8mL
3. 12mL
4. 16mL

Q3. Calcium phosphate is present in tooth enamel. Its nature is:

1. Basic.
2. Acidic.
3. Neutral.
4. Amphoteric.

Q4. Which of the following salts will give an aqueous solution having pH of almost 7?

1.  $NH_4NO_3$
2.  $NH_4Cl$
3.  $CaCl_2$



# Acids Bases And Salts Guide Answers

**R P Manchanda**



## **Acids Bases And Salts Guide Answers:**

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