

Acids and Bases

Multiple Choice: Acid-Base Theories

In the space provided, indicate which of the theories proposed best best explains each characteristic. Some answers may require more than one theory.

1. All dissolved LiOH is fully ionized.
a. not applicable
b. both a & b are applicable
c. **Arrhenius**
d. All of the above
2. Which of the following substances is both a Brønsted-Lowry base and an Arrhenius base? Explain your answer by describing the reaction.
a. H_2SO_4
b. **H_2SO_3**
c. H_2SO_2
d. H_2SO_3
3. In the following reaction, which substance acts as a Brønsted-Lowry acid? Explain your answer.
a. **H_2O**
b. H_2O
c. H_2O^+
d. OH^-
4. In the following reaction, which substance acts as a Brønsted-Lowry base? Explain your answer.
a. H_2O
b. **H_2O^+**
c. H_2O
d. OH^-
5. Which of the following is a polyprotic acid?
a. HNO_3
b. **H_2SO_4**
c. H_2SO_3
d. HCl

Acids Base Theories Answers

Roman Wölfel



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