

CBSE Sample Paper Class 12 Chemistry Set 1

Time allowed: 3 Hours

Max. Marks: 70

General Instructions:

- (a) All questions are compulsory
- (b) Section A: Q. no. 1 to 5 are very short questions and carry 1 mark each.
- (c) Section B: Q. no. 6 to 12 are short answer questions and carry 2 marks each.
- (d) Section C: Q. no. 13 to 24 are also short answer questions and carry 3 marks each.
- (e) Section D: Q. no. 25 to 27 are long answer questions and carry 5 marks each.
- (f) There is no overall choice. However an internal choice has been provided in two questions of one mark, two questions of two marks, four questions of three marks and all the three questions of five marks weightage. You have to attempt only one of the choices in such questions.
- (g) Use of log tables if necessary, use of calculators is not allowed.

Section-A (1 mark each)

Q1. What is the Normality of 1.5 M H_2SO_4 ?

OR

Why Al cannot be obtained by electrolysis of fused AlCl_3 ?

Q2. A reaction is found to be zero order. Will its molecularity be also zero.

Q3. Which out of Molarity or Molality varies with temperature?

OR

Define Adsorption.

Q4. Which method is used to concentrate the sulphide ores?

Q5. Name the elements of the 13th group and give general electronic configuration of their valence shells.

Section-B (2 marks each)

Q6. In a close packing of N spheres, how many (i) tetrahedral and (ii) octahedral sites are present?

OR

What is the role of salt-bridge in an electrochemical cell?

Q7. Mention any four factors affecting rate of a reaction.

OR

Describe the two uses of each of the following:

- (a) Copper sulphate (b) Silver nitrate
 - (i) For copper plating
 - (ii) As germicide and fungicide
- (b) Uses of Silver nitrate

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JagranJosh



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Table of Contents 12th Class Chemistry Solve Paper 2rbse

1. Understanding the eBook 12th Class Chemistry Solve Paper 2rbse
 - The Rise of Digital Reading 12th Class Chemistry Solve Paper 2rbse
 - Advantages of eBooks Over Traditional Books
2. Identifying 12th Class Chemistry Solve Paper 2rbse
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 12th Class Chemistry Solve Paper 2rbse
 - User-Friendly Interface
4. Exploring eBook Recommendations from 12th Class Chemistry Solve Paper 2rbse
 - Personalized Recommendations
 - 12th Class Chemistry Solve Paper 2rbse User Reviews and Ratings
 - 12th Class Chemistry Solve Paper 2rbse and Bestseller Lists

5. Accessing 12th Class Chemistry Solve Paper 2rbse Free and Paid eBooks
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 - 12th Class Chemistry Solve Paper 2rbse Budget-Friendly Options
6. Navigating 12th Class Chemistry Solve Paper 2rbse eBook Formats
 - ePub, PDF, MOBI, and More
 - 12th Class Chemistry Solve Paper 2rbse Compatibility with Devices
 - 12th Class Chemistry Solve Paper 2rbse Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 12th Class Chemistry Solve Paper 2rbse
 - Highlighting and Note-Taking 12th Class Chemistry Solve Paper 2rbse
 - Interactive Elements 12th Class Chemistry Solve Paper 2rbse
8. Staying Engaged with 12th Class Chemistry Solve Paper 2rbse
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 12th Class Chemistry Solve Paper 2rbse
9. Balancing eBooks and Physical Books 12th Class Chemistry Solve Paper 2rbse
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 12th Class Chemistry Solve Paper 2rbse
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine 12th Class Chemistry Solve Paper 2rbse
 - Setting Reading Goals 12th Class Chemistry Solve Paper 2rbse
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of 12th Class Chemistry Solve Paper 2rbse
 - Fact-Checking eBook Content of 12th Class Chemistry Solve Paper 2rbse
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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