

Define a pure substance

How can you distinguish a pure substance from a mixture?

What will happen to a pure substance if you heat it into the air?

What is chromatography used for?

Remember the three words on the diagram to help



Separate a blue ink into its constituent parts of color

Complete the word equations to calculate the RF value

$R_F =$

Remember knowing the R_F value allows you to identify a substance?

What is a precipitate?

Give some everyday examples of when precipitates are used

What is an alloy? Name the correct answer

- a. Mixture of two non-metals
- b. Mixture of two metals
- c. Mixture of two metalloids

What are the advantages of alloys?

What are the two types of chromatography?

Thin layer chromatography uses _____ It separates out all the _____ of a mixture. Paper chromatography is done on _____ of aluminium foil using _____ Thin layer chromatography is much more useful than paper as they are easier to use

What is the R_F value of the following chromatogram?

If the distance moved by substance B is 30mm and the distance moved by substance A is 10mm

What are the following separation techniques?



What separation technique would you use to separate an aqueous solution into pure?

How can salt be extracted using the process of crystallisation?

How can water not be separated by using a porous filter?

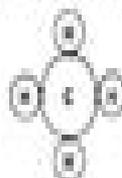
Remember a filter paper has to collect salt from water salt

Calculate the formula mass. Show your working out at 100, 10 x 16, 10 x 16, 10 x 16

10×16
 $10 \times 16 \times 10$, 10×16

Draw the following dot and cross diagrams

Remember draw dot and cross for the following full shells, 2, 8



What is the a diagram of? Oxygen



Explain why the molecular shape can affect the density of a gas

How is a blue ink solution separated?



What is the structure?



How many carbon atoms are there?

Explain how a single layer of graphite

Why is the material very strong?

What is the product used?

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