

## Chemical Reactions

# Intro. to Chemical Reactions

Name: \_\_\_\_\_  
 Date: \_\_\_\_\_  
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### Electrolysis Reaction

1. Write the chemical formula following compounds:

a) hydrogen gas:  $H_2$                       b) sodium phosphate:  $Na_3PO_4$   
 c) magnesium phosphate:  $Mg_3(PO_4)_2$                       d) sodium nitrate:  $NaNO_3$

2. Write the correct formula following compounds:

a)  $Ca(NO_3)_2$ : calcium nitrate                      b)  $FeCl_3$ : iron(III) chloride  
 c)  $CaSO_4$ : calcium sulfate                      d)  $NaF$ : sodium fluoride

### Chemical Equations: What they look like...

Here is what a chemical reaction looks like when it is written:



There are examples that provide hints to describe what happens during:

**Chemical equation** reacts with **reactants** and yields **products** and **physical states**

There are hints to solve stoichiometric problems: the numbers and the physical states. Here are a few problems:

### Critical Thinking Questions

1. Write the following reaction, with the correct state symbols.



**Answer:**  $FeCl_3(aq) + AgNO_3(aq) \rightarrow Fe(NO_3)_3(aq) + AgCl(s)$

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**Peggy S. Alier, Janet E. Snow, Bob Jones  
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