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## Hourly Calculations Worksheet ANSWER KEY

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1. Calculate the molar mass of each compound. Round your answer to the nearest 0.01 g/mol.

Answer Key:

$$\text{Mg} = 24.31 \text{ g/mol} + 2(15.99) \text{ g/mol} = 76.29 \text{ g/mol}$$

Answer Key:

$$\text{Mg} = 24.31 \text{ g/mol} + 1(16.00) \text{ g/mol} = 40.31 \text{ g/mol}$$

$$\text{Mg} = 24.31 \text{ g/mol} + 2(16.00) \text{ g/mol} = 56.31 \text{ g/mol}$$

2. Calculate the molar mass of each compound. Round your answer to the nearest 0.01 g/mol.

Answer Key:

$$\text{Mg} = 24.31 \text{ g/mol} + 1(16.00) \text{ g/mol} = 40.31 \text{ g/mol}$$

$$\text{Mg} = 24.31 \text{ g/mol} + 2(16.00) \text{ g/mol} = 56.31 \text{ g/mol}$$

Answer Key:

$$\text{Mg} = 24.31 \text{ g/mol} + 2(16.00) \text{ g/mol} = 56.31 \text{ g/mol}$$

3. Calculate the molar mass of each compound. Round your answer to the nearest 0.01 g/mol.

$$\text{Mg} = 24.31 \text{ g/mol} + 2(16.00) \text{ g/mol} = 56.31 \text{ g/mol}$$

4. Calculate the molar mass of each compound. Round your answer to the nearest 0.01 g/mol.

$$\text{Mg} = 24.31 \text{ g/mol} + 2(16.00) \text{ g/mol} = 56.31 \text{ g/mol}$$

5. Calculate the molar mass of each compound. Round your answer to the nearest 0.01 g/mol.

Answer Key:

$$\text{Mg} = 24.31 \text{ g/mol} + 1(16.00) \text{ g/mol} = 40.31 \text{ g/mol}$$

$$\text{Mg} = 24.31 \text{ g/mol} + 2(16.00) \text{ g/mol} = 56.31 \text{ g/mol}$$

Answer Key:

$$\text{Mg} = 24.31 \text{ g/mol} + 2(16.00) \text{ g/mol} = 56.31 \text{ g/mol}$$

6. Calculate the molar mass of each compound. Round your answer to the nearest 0.01 g/mol.

Answer Key:

$$\text{Mg} = 24.31 \text{ g/mol} + 2(16.00) \text{ g/mol} = 56.31 \text{ g/mol}$$

$$\text{Mg} = 24.31 \text{ g/mol} + 2(16.00) \text{ g/mol} = 56.31 \text{ g/mol}$$

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# Chemistry Unit 12 Answer Keys

**Emilie Sanchez**



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