

Chemistry – Unit 6 Worksheet 2

We have observed evidence that atoms of ^{12}C (graphite) are distributed over several particles based on their probability distribution. However, unlike molecular particles, each of these separately-observed atoms cannot be traced, when the particles are distributed, until particles are combined to form ^{12}C molecules. When the particles are distributed, we will see evidence that particles that form the other carbon isotopes (^{13}C and ^{14}C) are present in ratios that change as the time they live.

1. Write the formula and draw the particle diagram for each compound. The ratio of atoms in each compound is given.

Molecule molecular formula	12 particles 12 molecules	13 particles 12 molecules	14 particles 12 molecules	13 particles 12 molecules
	$^{12}\text{C}_2$	$^{13}\text{C}_2$	$^{14}\text{C}_2$	$^{13}\text{C}_2$

particle:



Diagram:

Molecule molecular formula	12 particles 12 molecules	13 particles 12 molecules	14 particles 12 molecules
	$^{12}\text{C}_3$	$^{13}\text{C}_3$	$^{14}\text{C}_3$

particle:



Diagram:

Molecule molecular formula	12 particles 12 molecules	13 particles 12 molecules	14 particles 12 molecules	13 particles 12 molecules
	$^{12}\text{C}_4$	$^{13}\text{C}_4$	$^{14}\text{C}_4$	$^{13}\text{C}_4$

particle:



Diagram:

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Ed Shewan, Garry Moes



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