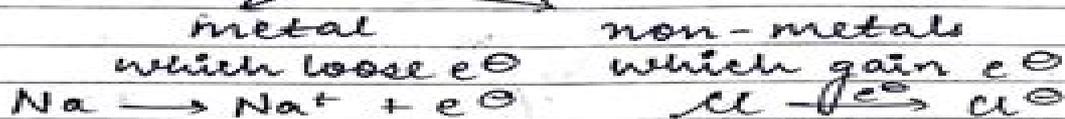


Periodic Table

Page No. 75

Lavoisier classification :-



cannot classify metalloids.

Prout's Hypothesis

All elements are made up of hydrogen.

$$At. wt. = n \times (At. wt. of H atom)$$

$$n = 1, 2, 3, 4, \dots$$

Drawback :- $Cl \rightarrow 35.5$ it is not whole no.

Dobereiner law of triads :-

3 elements \rightarrow Triad
 $[1, 2, 3]$

$$w_2 = \frac{w_1 + w_3}{2}$$

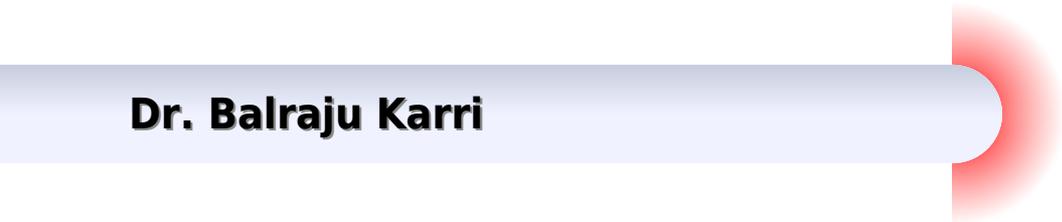
Drawback: A large no. of elements can't be classified in triads.

eg	Li	7		$\frac{7 + 39}{2}$	= 23
	Na	23			
	K	39			
eg	Cl	35.5		$\frac{35.5 + 127}{2}$	= 81.5
	Br	80			
	I	127			

Fe + Co Ni X
 55 58.9 58.6

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