

General Chemistry Semester 2 Study Guide

Mr. Price - Fall 2018

1st Period: Monday, June 11th, 8:00am - 10:00am

2nd Period: Wednesday, June 13th, 11:00am - 1:00pm

3rd Period: Thursday, June 14th, 11:00am - 1:00pm

Use your resources wisely throughout the semester. You may not get to the end of the book because of a variety of circumstances that are beyond your control. Nevertheless, make every effort to complete the year of chemistry that you are enrolled in, please, and I would appreciate your feedback. Please contact your teacher, Mr. Price, for additional insight.

Make yourself a learning schedule of the material covered throughout the semester. This is not just about what you suggest, but a more consistent & rigorous schedule of your study that will make your studying more efficient. Use the completed schedule to identify whether you have more work that requires additional attention.

Be flexible, make any needed adjustments throughout the semester, and communicate.

Study Guide: Reviewable

Make sure you finish the following material necessary to be able to pass your chemistry course. This means that:

1. Periodic Table
2. Chemical Properties, Acids
3. Molar Calculations Chapter
4. Stoichiometry Chapter
5. Naming ionic & molecular compounds using a correct prefix, suffix, charge, & nomenclature
6. Unit conversions (know these well) for use often:
 - a. 10^3 mg = 1 g
 - b. 1000 mL = 1 L
 - c. 1000 g = 1 kg
7. Molarity problems (know how to calculate)

Excellent Reading Challenge

Below are 100 pages of additional material and resources that will be helpful to you. Make sure to finish all assignments and problems.

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| <ol style="list-style-type: none">1. Molecular Structure of Matter (Chap. 10) - 100 pgs.<ul style="list-style-type: none">a. Molecular Structure of Matter (100 pgs)b. Molecular Structure2. Molecular Mass (Chap. 11) - 100 pgs.<ul style="list-style-type: none">a. Molar Massb. Molar Mass3. Stoichiometry (Chap. 12) - 100 pgs.<ul style="list-style-type: none">a. Stoichiometry (100 pgs)b. Specific Heat4. Equilibrium (Chap. 13) - 100 pgs. | <ol style="list-style-type: none">5. Nomenclature (Chap. 14) - 100 pgs.<ul style="list-style-type: none">a. Nomenclatureb. Nomenclaturec. Nomenclatured. Nomenclature6. Chemical Properties (Chap. 15) - 100 pgs.<ul style="list-style-type: none">a. Chemical Propertiesb. Chemical Propertiesc. Chemical Properties7. Kinetics and Equilibrium (Chap. 16) - 100 pgs. |
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