

- 1 Out of  $Zn^{2+}$  and  $Co^{2+}$  which one is paramagnetic? Also give the reason (Z for Zn= 30 , Co= 27)
- 2 What is covalent hydride?
- 3 Why the position of electron cannot be predicted with certainty although it shows particle nature?
- 4 In the structure of diborane ,indicate the formation of "banana bond".
- 5 Position of hydrogen is not justified in periodic table. Comment
- 6 Calculate the volume of hydrogen gas evolved at STP by the action of 9.2g of Na on water. (At. Mass : Na=23u , H=1u , O=16u)

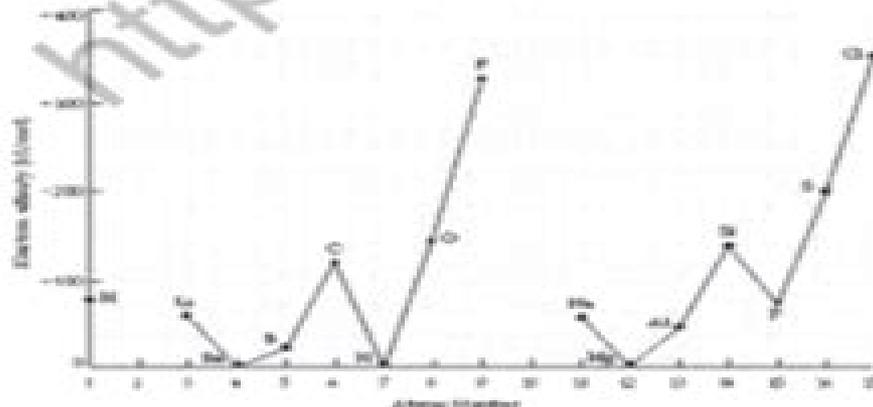
OR

- 7 What mass of  $CaCO_3$  is formed when 10g of  $CaO$  is heated with 20 g of air containing 25%  $CO_2$ . (At mass of Ca = 40u , C = 12u)
- 8 Calculate the number of moles of methanol ( $CH_3OH$ ) in 5 litres of its 2m solution if density of solution is 0.98 kg/L. Molar mass of methanol is 32g/mol
  - a) Give any one method of preparation of hydrogen peroxide.
  - b) Give an equation to show that hydrogen peroxide is an reducing agent.
- 9
  - a) Mention two contributions of green chemistry applicable in daily life.
  - b) Discuss the formation of ozone hole in brief( no reactions reqd.) .

10 State :

- a) Aufbau's Principle
- b) Pauli Exclusion principle with example

11



Answer the following questions based on the above graph: (Electron Affinity Vs Atomic no.):

- a) Why on moving from Li to Be , the value of electron gain enthalpy drops
- b) Cl has a higher peak than F. Explain

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