



Name: \_\_\_\_\_

Date: \_\_\_\_\_

## Chemistry – Study Guide

Ch 10 Study Guide Due: \_\_\_\_\_

Chapter 10 Test – \_\_\_\_\_

When studying for this test, use your 'do nows', notes, homework, class handouts, and your textbook.

### Vocabulary

Avogadro's number

Mole

molar mass

equivalence statement

conversion factor

*For all questions involving calculations, you must show all work, round your answers to the correct number of significant figures, and label your final answers with the correct units.*

### Measuring Matter - Chapter 10 – Section 10.1 – Pgs. 320-324

- Which of the following is NOT a representative particle?
  - Atom
  - Anion
  - Cation
  - All are representative particles
- Avogadro's number of representative particles is equal to one
  - Kilogram
  - Gram
  - Mole
  - Amu
- How many atoms of potassium does 1 mol of potassium contain?

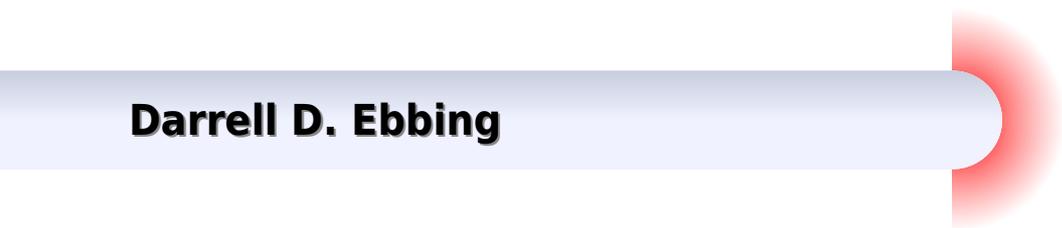
$$6.02 \times 10^{23} \text{ atoms}$$

- How many moles of xenon are equivalent to  $5.66 \times 10^{26}$  atoms?

$$\frac{5.66 \times 10^{26} \text{ atoms}}{1} \times \frac{1 \text{ mol}}{6.02 \times 10^{23} \text{ atoms}} = 940. \text{ mol Xe}$$

# Chemistry Matter And Change Chapter 10 Study Guide Answers

**Darrell D. Ebbing**



## **Chemistry Matter And Change Chapter 10 Study Guide Answers:**

*Chemistry, Student Study Guide* James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane  $\text{CH}_4$  from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement

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