

balance nuclear reaction equations

aluminum-26 undergoes positron emission
 potassium-40 undergoes beta decay (electron)
 uranium-238 undergoes alpha decay



electron configurations

Write the electron configuration for C.
 Write the electron configuration for Cr.
 Write the shorthand notation for Bi.



NOMENCLATURE (naming compounds, writing formulas)

Write the formula for each of the following compounds:

- | | |
|-----------------------------|---|
| a. Sulfur hexafluoride | SF_6 |
| b. Lithium nitride | Li_3N |
| c. Chromium (III) carbonate | $\text{Cr}_2(\text{CO}_3)_3$ |
| d. Tin (II) chloride | SnCl_2 |
| e. Ammonium acetate | $\text{NH}_4\text{C}_2\text{H}_3\text{O}_2$ (or $\text{CH}_3\text{COONH}_4$) |
| f. Mercury (I) chloride | Hg_2Cl_2 |
| g. Potassium bromate | KBrO_3 |
| h. Hydrosulfuric acid | $\text{H}_2\text{S(aq)}$ |
| i. Chloric acid | HClO_3 |
| j. Sulfurous acid | H_2SO_3 |

Name each of the following compounds:

- | | |
|------------------------------|-------------------------|
| a. CuSO_4 | Copper (II) sulfate |
| b. AlF_3 | aluminum fluoride |
| c. HI | hydrogen iodide |
| d. NO | nitrogen monoxide |
| e. $\text{H}_2\text{Se(aq)}$ | hydroselenic acid |
| f. HNO_3 | nitric acid |
| g. HNO_2 | nitrous acid |
| h. NaHSO_4 | Sodium hydrogen sulfate |

Half-life

Tritium (H-3) is a radioactive isotope of hydrogen with a half-life of 12.3 years. How long would it take for a 40.0 g sample to decay down to 1.25 g?



Fe-61 has a half-life of 6.00 min. Of a 100.0 mg sample, how much will remain after 18.0 min?



Average atomic mass

The element copper has naturally occurring isotopes with mass numbers of 63 and 65.

The relative abundance and atomic masses are 69.2% for a mass of 63amu and 30.8% for a mass of 65amu. Calculate the average atomic mass of copper.



Chemistry Lab 17 Answers

**Paul Levitz, Dennis O'Neil, Jack C.
Harris**



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Practical Chemistry Labs Leonard Saland,1989 Features self contained step by step activities using common materials and covering topics from food chemistry to papermaking and electrochemistry Illustrates the connection between the real world and chemistry concepts such as solutions chemistry acids and bases and more Includes teacher notes quizzes and answers to help monitor student progress The United States Catalog Mary Burnham,Carol Hurd,1928 *The Zinc and Iodine Book* Stephen DeMeo,2019-05-15 This book is for chemistry teachers who are thinking about reinventing their laboratory experiments that they provide to their students More than a collection of experiments it is an example of using a chemical theme to teach chemistry Instead of introducing many different chemicals per experiment as is the norm in most lab manuals this novel resource focuses on two commonly found elements Zinc and Iodine So what is so special about these elements At the heart of this resource is a colorful cyclic reaction between zinc and iodine one that produces a compound that can decompose back to its original elements This unique phenomenon demonstrates that matter not only changes but is also conserved through a chemical reaction Knowing that a compound can be the same but different than the reactants that formed it is to understand the essence of chemical change Complementing this reaction this book contains experimental activities that utilize the zinc and iodine theme to scaffold new concepts such as the properties of matter solid and gas stoichiometry equilibrium kinetics acid base chemistry and electrochemistry This teacher tested resource focuses on a set of safe substances that are appropriate for high school teachers who provide an advanced chemistry placement course and for college instructors teaching a first year chemistry laboratory sequence DC Super Stars #17 Facsimile Edition (2020-) #1 Paul Levitz,Dennis O'Neil,Jack C. Harris,2020-04-08 This 1977 issue features the debut of Huntress the untold origin of Green Arrow and an early adventure of the Legion of Super Heroes Offered to coincide with the Catwoman 80th Anniversary 100 Page Super Spectacular 1 *The Laboratory Digest* ,1953 **180 Practice Drills for the LSAT: Over 5,000 Questions to Build Essential LSAT Skills** Kaplan Test Prep,2023-06 180 Practice Drills for the LSAT includes over 5 000 questions to help you practice the skills you need to improve your score Every LSAT question tests skills in combination When you get a question wrong how do you pinpoint which of those skills was lacking in your performance This LSAT prep book takes the guesswork out of that analysis by testing each skill individually Whether you re at the beginning of your LSAT preparation or you re a seasoned LSAT veteran the skills that are tested here are the building blocks of score movement In addition to thousands of questions across 180 drills the book also includes Cheat Sheets of the must knows for every question and game type Comprehensive review guides to build fundamental skills in Logical Reasoning Reading Comprehension and Logic Games A crash course in our lexicon and approach for students who have prepped differently Planning resources to get the most out of your PrepTests Chemical Abstracts ,1910 **The Cumulative Book Index** ,1930 *Chemistry* Eugene LeMay, Jr.,Herbert Beall,Karen M. Robblee,Douglas C. Brower,Prentice-Hall Staff,2002-02 Gradwohl Laboratory Digest

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