

Writing Electron Configuration & Orbital Diagram

1. Write the long-form electron configuration of the following elements?

- i. Lithium _____
- ii. Magnesium _____
- iii. Potassium _____
- iv. Nickel _____
- v. Bromine _____

2. Write the abbreviated electron configuration of the following elements?

- i. Nitrogen _____
- ii. Argon _____
- iii. Manganese _____
- iv. Silver _____
- v. Barium _____

3. Identify the elements from the orbital filling diagrams.

Orbital Filling Diagram	Element
<p style="text-align: center;"> $\uparrow\downarrow$ 1s $\uparrow\downarrow$ 2s $\uparrow\downarrow$ \uparrow \square </p>	
<p style="text-align: center;"> $\uparrow\downarrow$ 1s $\uparrow\downarrow$ 2s $\uparrow\downarrow$ \uparrow \square \square </p>	
<p style="text-align: center;"> $\uparrow\downarrow$ 1s $\uparrow\downarrow$ 2s $\uparrow\downarrow$ \uparrow </p>	
<p style="text-align: center;"> $\uparrow\downarrow$ 1s $\uparrow\downarrow$ 2s $\uparrow\downarrow$ \uparrow \square </p>	
<p style="text-align: center;"> $\uparrow\downarrow$ 1s $\uparrow\downarrow$ 2s $\uparrow\downarrow$ \uparrow \square \square </p>	

Chemistry Electron Configuration Webquest Answer Key

Rachel Turoscy



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The Answer Key: A Comprehensive Explanation of Problem Solving Methods for General Chemistry Success (Volume One) (First Edition Rachel Turoscy, 2018-08-09 The Answer Key A Comprehensive Explanation of Problem Solving Methods for General Chemistry Success Volume 1 is a concise and accessible textbook that covers the critical information a student needs to understand the basic mathematics used in chemistry courses The book provides easy to understand step by step instructions for solving general chemistry problems The book begins with chapters dedicated to problem solving methodology and unit conversions In subsequent chapters the text covers important topics like ionic and covalent bonding chemical formula calculations solubility and reactions in aqueous solution gases the first law of Thermodynamics Quantum theory and electron configuration It also covers periodic trends the Lewis Dot Structures and bonding theories Each chapter contains sample problems and practice problems to help further understanding of how math and chemistry go hand in hand The Answer Key is an excellent resource for any undergraduate course that deals with the basic concepts of general chemistry

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