

Practice Problems

Answers: Answers are given at the end of these problems.

SECTION 18.1 PROPERTIES OF SOLUTIONS

1. The volume of the solution is 100 mL. The mass of the solution is 100 g. The mass of the solute is 10 g. The mass of the solvent is 90 g.

ANSWER 1:

2. The volume of the solution is 100 mL. The mass of the solution is 100 g. The mass of the solute is 10 g. The mass of the solvent is 90 g.

ANSWER 2:

3. The volume of the solution is 100 mL. The mass of the solution is 100 g. The mass of the solute is 10 g. The mass of the solvent is 90 g.

SECTION 18.2 CONCENTRATIONS OF SOLUTIONS

1. The volume of the solution is 100 mL. The mass of the solution is 100 g. The mass of the solute is 10 g. The mass of the solvent is 90 g.

ANSWER 1:

2. The volume of the solution is 100 mL. The mass of the solution is 100 g. The mass of the solute is 10 g. The mass of the solvent is 90 g.

ANSWER 2:

3. The volume of the solution is 100 mL. The mass of the solution is 100 g. The mass of the solute is 10 g. The mass of the solvent is 90 g.

- a. The volume of the solution is 100 mL. The mass of the solution is 100 g. The mass of the solute is 10 g. The mass of the solvent is 90 g.

ANSWER 3:

- b. The volume of the solution is 100 mL. The mass of the solution is 100 g. The mass of the solute is 10 g. The mass of the solvent is 90 g.

ANSWER 4:

4. The volume of the solution is 100 mL. The mass of the solution is 100 g. The mass of the solute is 10 g. The mass of the solvent is 90 g.

ANSWER 5:

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Paul A. Bartlett, Judith G. Koch



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