



## Computing

## COMP1

Unit 1 Problem Solving, Programming, Data Representation and Practical Exercise

Friday 25 May 2012 9.00 am to 11.00 am

**You will need to:**

- access the Electronic Answer Document
- refer to the Preliminary Material and the Skeleton Program

You must not use a calculator.

**Time allowed**

- 2 hours

**Instructions**

- Type the information required on the front of your Electronic Answer Document.
- Enter your answers into the Electronic Answer Document.
- Answer all questions.
- You will need access to:
  - a computer
  - a printer
  - appropriate software
  - an electronic version of the Skeleton Program
- Before the start of the examination make sure your **Centre Number, Candidate Name and Number** are shown clearly in the footer of every page of your Electronic Answer Document (not the front cover).

**Information**

- The marks for questions are shown in brackets.
- The maximum mark for this paper is 100.
- No extra time is allowed for printing and collating.
- The question paper is divided into four sections.  
You are advised to spend time on each section as follows:
  - Section A – 35 minutes
  - Section B – 20 minutes
  - Section C – 15 minutes
  - Section D – 50 minutes.

**At the end of the examination**

- Tie together all your printed Electronic Answer Document pages and hand them to the invigilator.

**Warning**

- It may not be possible to issue a result for this unit if your details are not on every page of the Electronic Answer Document.

# Aqa Computing Comp1 Electronic Answer Document 2014

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## **Aqa Computing Comp1 Electronic Answer Document 2014:**

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web the structure of an organic compound which on oxidation gives an acid that gives a single mono substituted product on nitration with  $\text{HNO}_3$  and  $\text{H}_2\text{SO}_4$  is medium view solution

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web jan 23 2023 exhaustive oxidation of organic molecules by  $\text{KMnO}_4$  will proceed until the formation of carboxylic acids therefore alcohols will be oxidized to carbonyls aldehydes and ketones and aldehydes and some ketones as in 3 above will be oxidized to carboxylic acids

**oxidation by  $\text{KMnO}_4$  researchgate** - Oct 05 2022

web an improved kinetic model for the high temperature oxidation of toluene has been developed using previously established reaction mechanisms for benzene and toluene

**toluene oxidation process and proper mechanism over  $\text{Co}_3$**  - May 12 2023

web oct 1 2020 in situ drifts combined with ptr tof ms quasi in situ xps and uv vis drs were introduced to learn the process of toluene oxidation confirmed that the reaction mechanism over catalyst followed the mars van krevelen mechanism and surface lattice oxygen played an important role in deeper oxidation toluene

mechanism of the oxidation of alcohols with  $\text{KMnO}_4$  - Aug 03 2022

web mechanism of the oxidation of alcohols with  $\text{KMnO}_4$  ask question asked 7 years 4 months ago modified 7 years 4 months ago viewed 29k times 10 many oxidising agents like chromate dichromate iodine in naoh n a o h etc seem to work via ester formation and elimination

**toluene 1 oxidation with  $\text{KMnO}_4$  mechanism 2 benzoic acid** - Jun 13 2023

web may 15 2020 toluene 1 oxidation with  $\text{KMnO}_4$  mechanism 2 benzoic acid  $\text{C}_6\text{H}_5\text{COOH}$   $\text{SOCl}_2$  3 benzoyl chloride  $\text{C}_6\text{H}_5\text{COCl}$   $\text{LiAlH}_4$   $\text{CH}_3\text{CHO}$  4 benzaldehyde  $\text{C}_6\text{H}_5\text{CHO}$   $\text{CH}_2\text{OHCH}_2\text{OH}$  h step 1 toluene oxidation with  $\text{KMnO}_4$   $\text{KMnO}_4$  is a strong oxidising agent which oxidise toluene to benzaldehyde in 1st step

**oxidation of alkylarenes to the corresponding acids using** - Dec 07 2022

web may 1 2004 oxidation of toluene using aqueous potassium permanganate was studied under heterogeneous condition in the presence of hydrodynamic cavitation and compared with the results of the reaction under acoustic cavitation

**can acidified or neutral  $\text{KMnO}_4$  oxidise toluene to benzoic acid** - Jan 08 2023

web oct 15 2014 1 answer sorted by 6 here are the three equations describing the reduction of manganese and concurrent oxidation of whatever substrate may be present under basic neutral and acidic conditions respectively  $\text{Mn}^{7+} + 4\text{e}^- + \text{H}_2\text{O} \rightarrow \text{Mn}^{3+} + 2\text{OH}^-$   $\text{Mn}^{7+} + 3\text{e}^- + 4\text{H}^+ \rightarrow \text{Mn}^{4+} + 2\text{H}_2\text{O}$   $\text{Mn}^{7+} + 5\text{e}^- + 8\text{H}^+ \rightarrow \text{Mn}^{2+} + 4\text{H}_2\text{O}$  basic

**what is the mechanism of toluene oxidation by  $\text{KMnO}_4$  reddit** - Jul 14 2023

web mar 15 2019 what is the mechanism of toluene oxidation by  $\text{KMnO}_4$  it is quite simple reaction that  $\text{KMnO}_4$  can oxidize benzylic hydrogen in toluene and synthesize benzoic acid however i cannot find exact mechanism of reaction in

**oxidation of aromatic alkanes with  $\text{KMnO}_4$  to give carboxylic acids** - Feb 09 2023

web description treatment of an alkylbenzene with potassium permanganate results in oxidation to give the benzoic acid notes the position directly adjacent to an aromatic group is called the benzylic position the reaction only works if there is a hydrogen attached to the carbon examples

**oxidation of alkylarenes to the corresponding acids using** - Mar 10 2023

web may 1 2004 the oxidation of toluene by aq  $\text{KMnO}_4$  under hydrodynamic cavitation was taken as a model reaction and

various parameters have been optimized the oxidation of toluene by aq  $\text{KMnO}_4$  gives benzoic acid scheme 1  $\text{KMnO}_4$  in turn is reduced to  $\text{MnO}_2$  the reaction did not give any other byproduct