

Name: _____ ANSWER KEY _____ Class: _____ Date: _____

Unique Properties of Water!

Word Bank:

Adhesion	Evaporation	Polar	Surface tension
Cohesion	Freezing	Positive	Universal solvent
Condensation	Melting	Sublimation	
Dissolve	Negative		

1. The electrons are not shared equally between the hydrogen and oxygen atoms of water creating a Polar molecule.
2. The polarity of water allows it to dissolve most substances. Because of this it is referred to as the universal solvent.
3. Water molecules stick to other water molecules. This property is called cohesion.
4. Hydrogen bonds form between adjacent water molecules because the positive charged hydrogen end of one water molecule attracts the negative charged oxygen end of another water molecule.
5. Water molecules stick to other materials due to its polar nature. This property is called adhesion.
6. Hydrogen bonds hold water molecules closely together which causes water to have high surface tension. This is why water tends to clump together to form drops rather than spread out into a thin film.
7. Condensation is when water changes from a gas to a liquid.
8. Sublimation is when water changes from a solid directly to a gas.
9. Freezing is when water changes from a liquid to a solid.
10. Melting is when water changes from a solid to a liquid.
11. Evaporation is when water changes from a liquid to a gas.
12. Why does ice float?
Water expands as it freezes, so it is LESS DENSE AS A SOLID.
13. What property refers to water molecules resembling magnets? How are these alike?
Polar bonds create positive and negative ends of the molecule. The positive ends are attracted to other negative ends, creating hydrogen bonds between molecules.

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Table of Contents Biology Section 3 Answer Key Water

1. Understanding the eBook Biology Section 3 Answer Key Water
 - The Rise of Digital Reading Biology Section 3 Answer Key Water
 - Advantages of eBooks Over Traditional Books
2. Identifying Biology Section 3 Answer Key Water
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Biology Section 3 Answer Key Water
 - User-Friendly Interface
4. Exploring eBook Recommendations from Biology Section 3 Answer Key Water
 - Personalized Recommendations
 - Biology Section 3 Answer Key Water User Reviews and Ratings
 - Biology Section 3 Answer Key Water and Bestseller Lists
5. Accessing Biology Section 3 Answer Key Water Free and Paid eBooks
 - Biology Section 3 Answer Key Water Public Domain eBooks
 - Biology Section 3 Answer Key Water eBook Subscription Services

- Biology Section 3 Answer Key Water Budget-Friendly Options
- 6. Navigating Biology Section 3 Answer Key Water eBook Formats
 - ePub, PDF, MOBI, and More
 - Biology Section 3 Answer Key Water Compatibility with Devices
 - Biology Section 3 Answer Key Water Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Biology Section 3 Answer Key Water
 - Highlighting and Note-Taking Biology Section 3 Answer Key Water
 - Interactive Elements Biology Section 3 Answer Key Water
- 8. Staying Engaged with Biology Section 3 Answer Key Water
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Biology Section 3 Answer Key Water
- 9. Balancing eBooks and Physical Books Biology Section 3 Answer Key Water
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Biology Section 3 Answer Key Water
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Biology Section 3 Answer Key Water
 - Setting Reading Goals Biology Section 3 Answer Key Water
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Biology Section 3 Answer Key Water
 - Fact-Checking eBook Content of Biology Section 3 Answer Key Water
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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