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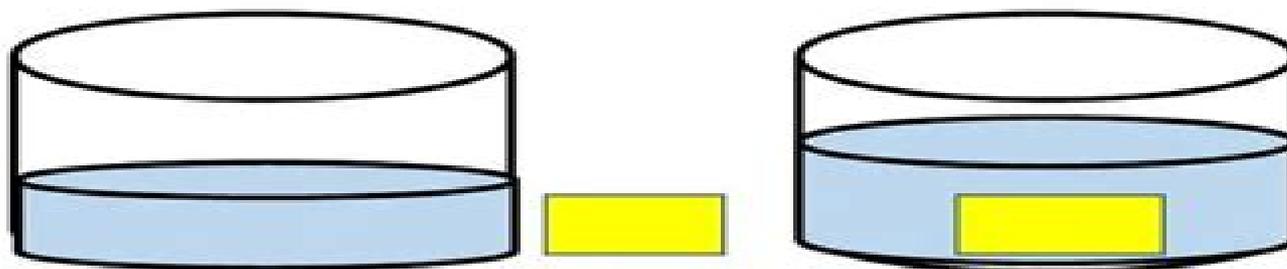
PBS Density by Displacement Simulation

Density is the compactness of matter. Density describes **the quantity of matter (mass) per unit of volume (3-dimensional space)**. Density is calculated as the mass of the matter divided by volume.

$$\rho = \frac{m}{V}$$

The symbol for density is Greek letter rho (ρ). The units for mass are in grams (g) and the units for volume is cubic centimeter (cm^3). It should be noted that 1 cm^3 is equal to 1 ml of volume. $1 \text{ cm}^3 = 1 \text{ ml}$. The units for density are g/cm^3 .

Density is calculated at mass divided by volume. You will determine density of solid materials. The volume will be determined by the displacement method. In the displacement method, the object is totally submerged within the liquid. The level or volume of the liquid inside of the container will increase because the new volume after the object is added will include the object's volume.



Volume of liquid without the submerged solid object = 200 ml (200 cm^3).

Volume of liquid with the submerged solid object = 265 ml (265 cm^3). The new volume is the volume of the liquid and volume of the object together.

The volume of the solid object is determined as the difference between the combined volume (liquid + solid) and the original volume of the liquid.

Volume object = $265 \text{ cm}^3 - 200 \text{ cm}^3 = 65 \text{ cm}^3$.

Density Lab Activities

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