

Unit Test: Chemical Bonding (A) **ANSWER KEY**

100

Part I: Matching (2 points each) Please match the term with the correct definition. There will be extra choices and some choices may be used more than once.

- | | | |
|----------------------|-------------------------|----------------------|
| A. Electronegativity | F. Molecular Formula | K. Crystal |
| B. Covalent bonding | G. Linear | L. Pyramid |
| C. Ductile | H. Non-polar | M. Polar |
| D. Aqueous | I. Solid | N. VSEPR |
| E. Tetrahedral | J. Trigonal bipyramidal | O. Malleable |
| P. Octet Rule | Q. Ionic bonding | R. Empirical Formula |

1. K A 3-D, repeating pattern of + and - ions, formed by ionic compound
2. A Tendency for an atom to attract electrons when chemically bonded to another atom
3. B A sharing of electrons
4. P Atoms will gain or lose enough electrons in order to become isoelectronic with a noble gas
5. Q A transfer of electrons
6. R A chemical formula that is arranged in the smallest whole number ratio
7. D The term that means dissolved in water
8. F A chemical formula that describes the makeup of a single molecule
9. J The shape (geometry) that is always an exception to the octet rule
10. M A bond where electrons are shared unequally between atoms
11. L One of the shapes (geometries) that will always be polar
12. H A bond where electrons are shared equally between atoms

Part II: Multiple Choice (2 points each) Please circle the most correct answer. Write your answer on the given lines.

1. E A triple covalent bond shares _____ electrons between atoms.

a. two	d. five
b. three	e. six
c. four	f. Unable to tell
2. C Which of the following sets of atoms can be joined by a covalent bond?

a. Be and F	d. Mg and Na
b. Na and Cl	e. Li and Cl
c. S and F	f. None of these

Bonding Test Answer Key

V A Coveney



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