

Average Atomic Mass Worksheet

Answer the following questions.

1. Look at the data presented below about sulfur

Sulfur-32 → 95.0% abundance
Sulfur-33 → 0.76% abundance
Sulfur-34 → 4.22% abundance
Sulfur-36 → 0.014% abundance

Using the information provided, determine the average atomic mass of sulfur.

2. An element has three isotopes, two of which are as follows

Isotope 1 → 42 amu (75% abundance)
Isotope 2 → 44 amu (10% abundance)

If the 3rd isotope has a mass of 45 amu, determine the average atomic mass of this element.

3. Chlorine has two isotopes, the information about which is presented below

(Isotope 1) Mass → 35 amu with an abundance of 75.53%
(Isotope 2) Mass → 37 amu with an abundance of 24.47%

Based on this data, what is the average atomic mass of chlorine?

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Yan Bai



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