

Chapter 7 - Decoding Chemical Formulas

As you briefly mentioned in a previous lesson, the chemical composition of a given compound is usually represented by a chemical formula.

- The formula indicates the constituent elements and the number of atoms of each element written as a subscript. This is the "atomic per formula unit," or apfu.
- For example, gold uses the number-19, oxygen uses-8, and silver uses-107.
- Each element abbreviation starts with a capital letter, and if a second letter is present it will always be lowercase.
- These letters are combined numerically together into a single formula with no spaces and no special signs and symbols typical of other scientific fields.

Consequently, a formula is simply "19" and then one silver atom per formula unit.

Chemical copper (200) from a simple formula is with "63.5", consisting of 1 silver atom and 1 copper-atom per formula unit.

Hydrochloric acid is HCl (page 127) has the formula "H1Cl1", consisting of 1 atom of H and 1 atom of Cl per formula unit.

Sometimes with an example, you can read formulas of the form "X₂Y₃Z₄" as "the given formula which belongs to the others "group of elements" (page 200), "H₂O", "Fe₂SO₄".

In that case there is 2 H, 1 O, and 1 S atom per formula unit. This is called a unit formula (page 10) and follows the other way of the world except up to 2 atoms per formula unit.

The Phases and Measures of Charge Neutrality

All compounds, including all minerals, must be charge neutral.

This means that every positive charge must be balanced by a negative charge. Hence that cations have a negative charge and anions have a positive charge.

Identifying charge neutrality

1. Quartz, SiO₂, is composed of 1 Si cation with a charge of +4 and two O anions with a -2 charge.
2. Pyrite, FeS₂, is composed of one Fe (+2) cation, and two (-1) anions and that 2(+2) = 0 atoms.

The concept of charge neutrality is important for understanding mineral substitution of ions and their chemical properties, and will be discussed later in this lecture.

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Sandra Mirabelli



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Table of Contents Decoding Chemical Formulas Answer Key

1. Understanding the eBook Decoding Chemical Formulas Answer Key
 - The Rise of Digital Reading Decoding Chemical Formulas Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Decoding Chemical Formulas Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Decoding Chemical Formulas Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Decoding Chemical Formulas Answer Key
 - Personalized Recommendations
 - Decoding Chemical Formulas Answer Key User Reviews and Ratings
 - Decoding Chemical Formulas Answer Key and Bestseller Lists
5. Accessing Decoding Chemical Formulas Answer Key Free and Paid eBooks
 - Decoding Chemical Formulas Answer Key Public Domain eBooks
 - Decoding Chemical Formulas Answer Key eBook Subscription Services
 - Decoding Chemical Formulas Answer Key Budget-Friendly Options

6. Navigating Decoding Chemical Formulas Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Decoding Chemical Formulas Answer Key Compatibility with Devices
 - Decoding Chemical Formulas Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Decoding Chemical Formulas Answer Key
 - Highlighting and Note-Taking Decoding Chemical Formulas Answer Key
 - Interactive Elements Decoding Chemical Formulas Answer Key
8. Staying Engaged with Decoding Chemical Formulas Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Decoding Chemical Formulas Answer Key
9. Balancing eBooks and Physical Books Decoding Chemical Formulas Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Decoding Chemical Formulas Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Decoding Chemical Formulas Answer Key
 - Setting Reading Goals Decoding Chemical Formulas Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Decoding Chemical Formulas Answer Key
 - Fact-Checking eBook Content of Decoding Chemical Formulas Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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