

Name: \_\_\_\_\_

Date: \_\_\_\_\_

**Colligative Properties Worksheet**

1. For a 0.222 m aqueous solution of sucrose,  $C_{12}H_{22}O_{11}$ . What is the boiling point?
2. What is the normal boiling point of a 0.122 m aqueous solution of KI?
3. Antifreeze and freeze consists of ethylene glycol,  $C_2H_6O_2$ , a nonvolatile nonelectrolyte. Calculate the boiling point and freezing point of a 25.0 mass percent solution of ethylene glycol in water.
4. Calculate the freezing point of a solution containing 42.0 g of eucalyptol ( $C_{15}H_{26}O$ ), a fragrant substance found in the leaves of eucalyptus tree, dissolved in a 0.600 kg solution with  $CHCl_3$ .  $K_f$  for  $CHCl_3$  is  $-4.68\text{ }^\circ\text{C}/m$  and normal freezing point is  $-63.5\text{ }^\circ\text{C}$ .
5. List the following aqueous solutions in order of their expected freezing points: 0.050 m  $CaCl_2$ ; 0.15 m  $NaCl$ ; 0.10 m  $SO_2$ ; 0.10 m  $C_{12}H_{22}O_{11}$ , starting with the most depressed.
6. A solution of an unknown nonvolatile nonelectrolyte was prepared by dissolving 0.250 g in 40.0 g  $CCl_4$ . The normal boiling point of the resultant solution increased by  $0.357\text{ }^\circ\text{C}$ . Calculate the molar mass of the solute.  $K_b = 5.02\text{ }^\circ\text{C}/m$
7. Camphor,  $C_{10}H_{16}O$ , melts at  $179.8\text{ }^\circ\text{C}$ . It also has particularly large freezing point depression constant,  $K_f = 40\text{ }^\circ\text{C}/m$ . When 0.186 g of an organic substance of unknown molar mass is dissolved in 22.01 g of liquid camphor, the freezing point of the mixture is found to be  $176.7\text{ }^\circ\text{C}$ . What is the approximate molar mass of the solute?  $K_f = 40\text{ }^\circ\text{C}/m$

# Colligative Properties Worksheet Ch13 Part A Calculations

**L Cohen**



**Colligative Properties Worksheet Ch13 Part A Calculations:**

## Unveiling the Magic of Words: A Overview of "**Colligative Properties Worksheet Ch13 Part A Calculations**"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their capability to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**Colligative Properties Worksheet Ch13 Part A Calculations**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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