

Colligative Properties, Homework 13

Table 1: Constants

Name	Formula	$k_b \frac{^{\circ}\text{C}}{m_{\text{particles}}}$	$k_f \frac{^{\circ}\text{C}}{m_{\text{particles}}}$
Water	H ₂ O	0.512	1.86
Naphthalene	C ₁₀ H ₈	*****	6.9
Ethyl alcohol	C ₂ H ₆ O	1.22	1.99
Acetic acid	HC ₂ H ₃ O ₂	3.07	3.90
Benzene	C ₆ H ₆	2.53	4.90
Nitrobenzene	C ₆ H ₅ NO ₂	5.24	7.00

- How many particles are generated when each of the following compounds are dissolved in water. a) FeCl₂, b) NaCl, c) C₆H₁₂O₆ (glucose), d) Cr₂(SO₄)₃, e) CH₃NH₂ (methyl amine)
- Calculate the a) freezing point and b) boiling point when 78.75 g NaCl are dissolved in 275 g of water.
- Antifreeze in cars is usually made by mixing ethylene glycol, HOCH₂CH₂OH, with water to make a 50.0% ~~volume/volume~~ solution. If the specific gravity of ethylene glycol is 1.113, calculate a) the boiling and b) the freezing point of this solution. Assume that water is the solvent.
- Using the solution made in problem 3, calculate the a) k_b and b) k_f for ethylene glycol assuming it is the solvent.
- Calculate the molecular mass of ethylene glycol if a 150 g sample is placed into 250 g of ethyl alcohol, which has a boiling point of 78°C and the resulting mixture has a boiling point of 89.8°C.

Colligative Properties Homework Answer

Patrick Vollmar



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Table of Contents Colligative Properties Homework Answer

1. Understanding the eBook Colligative Properties Homework Answer
 - The Rise of Digital Reading Colligative Properties Homework Answer
 - Advantages of eBooks Over Traditional Books
2. Identifying Colligative Properties Homework Answer
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Colligative Properties Homework Answer
 - User-Friendly Interface
4. Exploring eBook Recommendations from Colligative Properties Homework Answer
 - Personalized Recommendations
 - Colligative Properties Homework Answer User Reviews and Ratings
 - Colligative Properties Homework Answer and Bestseller Lists
5. Accessing Colligative Properties Homework Answer Free and Paid eBooks
 - Colligative Properties Homework Answer Public Domain eBooks
 - Colligative Properties Homework Answer eBook Subscription Services

- Colligative Properties Homework Answer Budget-Friendly Options
- 6. Navigating Colligative Properties Homework Answer eBook Formats
 - ePub, PDF, MOBI, and More
 - Colligative Properties Homework Answer Compatibility with Devices
 - Colligative Properties Homework Answer Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Colligative Properties Homework Answer
 - Highlighting and Note-Taking Colligative Properties Homework Answer
 - Interactive Elements Colligative Properties Homework Answer
- 8. Staying Engaged with Colligative Properties Homework Answer
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Colligative Properties Homework Answer
- 9. Balancing eBooks and Physical Books Colligative Properties Homework Answer
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Colligative Properties Homework Answer
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Colligative Properties Homework Answer
 - Setting Reading Goals Colligative Properties Homework Answer
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Colligative Properties Homework Answer
 - Fact-Checking eBook Content of Colligative Properties Homework Answer
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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