

Chapter 2: Chemistry Comes Alive

Matter is anything that occupies space and has mass—states of matter are solid, liquid, and gas.

Energy

Energy is defined as the capacity to do work.

- **Kinetic energy** – energy in action
- **Potential energy** – stored energy; inactive energy that has potential

Forms of energy:

1. **Chemical energy** – form stored in the bonds of chemical substances
2. **Electrical energy** – results from movement of charged particles
3. **Mechanical energy** – energy directly involved in moving matter; legs moving pedals of bike
4. **Radiant energy/electromagnetic energy** – energy that travels in waves

Energy can be converted but some of the energy is lost into an unusable form of energy.

Example: Electrical energy is converted to light energy in a light bulb but some energy is lost to heat (why light bulbs become hot).

Similarly all energy conversion on the body liberate heat—helping to maintain our body temp and ensuring that chemical reactions occur (chemical reactions are more likely to happen in increased temps).

Atoms and Elements

Elements – all matter is composed of elements; unique substances that cannot be broken down into simpler substance by ordinary chemical methods

** Carbon, oxygen, hydrogen, and nitrogen make up about 96% of the body's weight.*

Atoms – all elements are composed of these more or less identical particles or building blocks; composed of nucleus and subatomic particles

Subatomic particle	Charge	Location	Mass
Protons	Positive	Nucleus	Aprox 1 amu
Neutrons	Neutral	Nucleus	Aprox 1 amu
Elections	Negative	Elections cloud, orbits around nucleus	Aprox 0 amu or 1/2000 amu

Elections orbit around the nucleus, and the **orbital model** depicts the probable regions of greatest electron density by density shading

Chapter 2 Chemistry Comes Alive Worksheet

Jessica J Manson



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