

CHAPTER 11 REVIEW*Gases***SECTION 2****SHORT ANSWER** Answer the following questions in the space provided.

1. State whether the pressure of a fixed mass of gas will increase, decrease, or stay the same in the following circumstances:

_____ a. temperature increases, volume stays the same

_____ b. volume increases, temperature stays the same

_____ c. temperature decreases, volume stays the same

_____ d. volume decreases, temperature stays the same

2. Two sealed flasks, A and B, contain two different gases of equal volume at the same temperature and pressure. Assume that flask A is warmed as flask B is cooled. Will the pressure in the two flasks remain equal? If not, which flask will have the higher pressure?

PROBLEMS Write the answer on the line to the left. Show all your work in the space provided.

3. A bicycle tire is inflated to 55 lb/in.² at 15°C. Assume that the volume of the tire does not change appreciably once it is inflated.

a. If the tire and the air inside it are heated to 30°C by road friction, does the pressure in the tire increase or decrease? (Assume the volume of air in the tire remains constant.)

b. Because the temperature has doubled, does the pressure double to 110 psi?

c. What will the pressure be when the temperature has doubled? Express your answer in pounds per square inch.

Chapter 1 review Gases Section Short Answer

Cecie Starr, Beverly McMillan



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Table of Contents Chapter 1review Gases Section Short Answer

1. Understanding the eBook Chapter 1review Gases Section Short Answer
 - The Rise of Digital Reading Chapter 1review Gases Section Short Answer
 - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 1review Gases Section Short Answer
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chapter 1review Gases Section Short Answer
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 1review Gases Section Short Answer
 - Personalized Recommendations
 - Chapter 1review Gases Section Short Answer User Reviews and Ratings
 - Chapter 1review Gases Section Short Answer and Bestseller Lists

5. Accessing Chapter 1review Gases Section Short Answer Free and Paid eBooks
 - Chapter 1review Gases Section Short Answer Public Domain eBooks
 - Chapter 1review Gases Section Short Answer eBook Subscription Services
 - Chapter 1review Gases Section Short Answer Budget-Friendly Options
6. Navigating Chapter 1review Gases Section Short Answer eBook Formats
 - ePub, PDF, MOBI, and More
 - Chapter 1review Gases Section Short Answer Compatibility with Devices
 - Chapter 1review Gases Section Short Answer Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chapter 1review Gases Section Short Answer
 - Highlighting and Note-Taking Chapter 1review Gases Section Short Answer
 - Interactive Elements Chapter 1review Gases Section Short Answer
8. Staying Engaged with Chapter 1review Gases Section Short Answer
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chapter 1review Gases Section Short Answer
9. Balancing eBooks and Physical Books Chapter 1review Gases Section Short Answer
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chapter 1review Gases Section Short Answer
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chapter 1review Gases Section Short Answer
 - Setting Reading Goals Chapter 1review Gases Section Short Answer
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chapter 1review Gases Section Short Answer
 - Fact-Checking eBook Content of Chapter 1review Gases Section Short Answer
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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