

AP Chemistry Chapter 20 Electrochemistry Practice Free Response #1



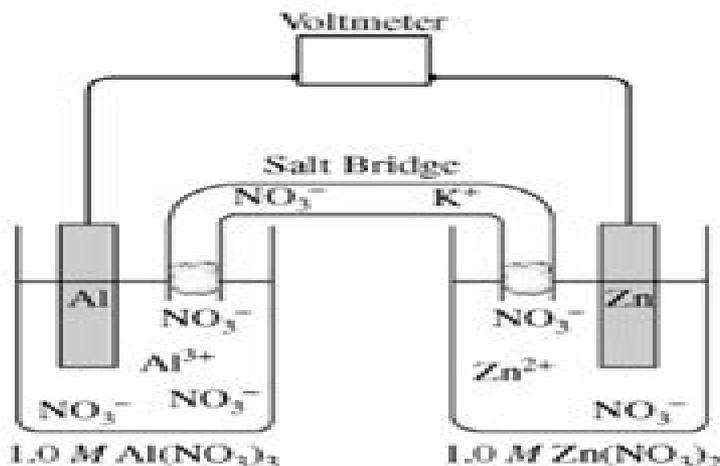
Respond to the following statements and questions that relate to the species and the reaction represented above.

(a) Write the complete electron configuration (e.g.,  $1s^2 2s^2 \dots$ ) for  $\text{Zn}^{2+}$ .

(b) Which species,  $\text{Zn}$  or  $\text{Zn}^{2+}$ , has the greater ionization energy? Justify your answer.

(c) Identify the species that is oxidized in the reaction.

The diagram shows a galvanic cell based on the reaction. Assume that the temperature is  $25^\circ\text{C}$ .



(d) The diagram includes a salt bridge that is filled with a saturated solution of  $\text{KNO}_3$ . Describe what happens in the salt bridge as the cell operates.

(e) Determine the value of the standard voltage,  $E^\circ$ , for the cell.

# Chapter 20 Electrochemistry Test

**JL Elias**



## **Chapter 20 Electrochemistry Test:**

## The Enigmatic Realm of **Chapter 20 Electrochemistry Test**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing short of extraordinary. Within the captivating pages of **Chapter 20 Electrochemistry Test** a literary masterpiece penned by way of a renowned author, readers set about a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting effect on the hearts and minds of those that partake in its reading experience.

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