

# Unit 1 study guide

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Class	Unit 1 - Chemistry of Life
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## Topic 1.1 : Structure of Water and Hydrogen Bonding

### 1. Structure /Polarity

- 2 elements : Oxygen and Hydrogen ( 1:2 ratio)
- Oxygen is more electronegative than hydrogen , resulting in an unequal sharing of electrons between oxygen and hydrogen. This is called Polarity, where there are differences in atoms electronegativity , a water molecule has polarity
- The bond is a covalent bond between the two elements ( they share electrons)
- The bond between other water molecules are hydrogen bond ( a weak bond between the negative and positive regions of two separate molecules).



### 2. Properties of water

#### LIVING THINGS DEPEND ON WATER'S PROPERTIES

- When two of the same molecules form hydrogen bonds with each other this is called cohesion ( water and water ) .
- When two different molecules form hydrogen bonds with each other this is called adhesion ( amino acid and water ) .

# Biology 1study Guide

**Jim Brinegar**



## **Biology 1study Guide:**

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pictures to help you see and remember the concepts and term In addition to technical diagrams and tables the pictures include many funny cartoons throughout the book to make learning the material more fun and memorable If you are frustrated by technical jargon and just want the important concepts explained in a straightforward way so you can do well in your biology course this is definitely the text for you

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