

Critical Thinking Questions

Answer CTQs 15 and 16 as a team.

15. In Model 3,

- How many moles of $\text{Mg}^{2+}(\text{aq})$ are present in beaker B?
- How many moles of $\text{OH}^{-}(\text{aq})$ are present in beaker A?
- What is the volume of the final solution in beaker C (where the contents of beakers A and B have been combined)?
- Assuming that no reaction occurs, what would be (Mg^{2+}) in beaker C?
- Assuming that no reaction occurs, what would be (OH^{-}) in beaker C?
- Write the expression for the reaction quotient, Q_{sp} , for this reaction:



Using your answers to parts d and e, calculate the value of the reaction quotient, Q_{sp} , in beaker C.

- Given the K_{sp} for $\text{Mg}(\text{OH})_2$ at $25\text{ }^{\circ}\text{C}$ is 1.80×10^{-11} , is the system in beaker C at equilibrium upon mixing? If not, in what direction will it shift?
- Does a precipitate of $\text{Mg}(\text{OH})_2(\text{s})$ form?

16. Using data from Model 3 and ideas from the Information section on page 258, explain why we can expect that there will be no $\text{NaNO}_3(\text{s})$ precipitate formed.

Answers To Chemactivity 15

Richard Samuel Moog



Answers To Chemactivity 15:

Chemistry Richard S. Moog, John J. Farrell, 2017-06-26 In the newly updated 7th Edition *Chemistry A Guided Inquiry* continues to follow the underlying principles developed by years of extensive research on how students learn and draws on testing by those using the POGIL methodology This text follows the principles of inquiry based learning and correspondingly emphasizes underlying chemistry concepts and the reasoning behind them This text provides an approach that follows modern cognitive learning principles by having students learn how to create knowledge based on experimental data and how to test that knowledge *Organic Chemistry, a Guided Inquiry* Andrei Straumanis, 2004 **Introductory Chemistry**

Michael P. Garoutte, Ashley B. Mahoney, 2015-08-10 The ChemActivities found in *Introductory Chemistry A Guided Inquiry* use the classroom guided inquiry approach and provide an excellent accompaniment to any one semester *Introductory text* Designed to support *Process Oriented Guided Inquiry Learning POGIL* these materials provide a variety of ways to promote a student focused active classroom that range from cooperative learning to active student participation in a more traditional setting **General, Organic, and Biological Chemistry** Michael P. Garoutte, 2014-02-24 Classroom activities to support a *General Organic and Biological Chemistry text* Students can follow a guided inquiry approach as they learn chemistry in the classroom *General Organic and Biological Chemistry A Guided Inquiry* serves as an accompaniment to a *GOB Chemistry text* It can suit the one or two semester course This supplemental text supports *Process Oriented Guided Inquiry Learning POGIL* which is a student focused group learning philosophy of instruction The materials offer ways to promote a student centered science classroom with activities The goal is for students to gain a greater understanding of chemistry through exploration

[Index of Conference Proceedings Received](#), 1979 **Department of Defense Appropriations for Fiscal Year 1993:**

Appendix submitted questions and answers United States. Congress. Senate. Committee on Appropriations.

Subcommittee on Defense, 1992 *Department of Defense Appropriations for Fiscal Year 1993* United States. Congress.

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Defense, 1993 **Index of Conference Proceedings Received, 1974-1978** British Library. Lending Division, 1980

Process Oriented Guided Inquiry Learning (POGIL) Richard Samuel Moog, 2008 POGIL is a student centered group learning pedagogy based on current learning theory This volume describes POGIL s theoretical basis its implementations in diverse environments and evaluation of student outcomes **Chemical Abstracts**, 2002 **Federation Proceedings**

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