

## Grade 12 Chemistry Exam Review

Below is a list of major topics covered on the exam for this course. If a topic is not listed, it will NOT be on the exam.

Your studying should begin with your class notes, assigned problems, textbook problems and extra review questions suggested below. This list is a guide and some specific questions on the exam may not appear here.

Good luck ☺

### Unit #1 – Energy Changes Text Reference: Chapter 5

- Energy changes in reactions (system, surroundings, endothermic vs. exothermic)
- Specific Heat Capacity
- Calorimetry,  $q=mc\Delta T$
- Molar enthalpies
- Representing Enthalpy Changes (Thermochemical Equations, Potential Energy Diagrams, etc.)
- Hess' Law
- Standard Enthalpies of Formation

### Sample Exam Questions

- Given the following information:

$$2\text{CH}_3\text{OH}(\text{l}) + 3\text{O}_2(\text{g}) \rightarrow 2\text{CO}_2(\text{g}) + 4\text{H}_2\text{O}(\text{l}) \quad \Delta H^\circ = -727 \text{ kJ}\cdot\text{mol}^{-1}$$

$$\Delta H_f^\circ \text{CO}_2(\text{g}) = -394 \text{ kJ}\cdot\text{mol}^{-1}, \quad \Delta H_f^\circ \text{H}_2\text{O}(\text{l}) = -286 \text{ kJ}\cdot\text{mol}^{-1}$$

What is the standard heat formation,  $\Delta H_f^\circ$  (kJ·mol<sup>-1</sup>) of CH<sub>3</sub>OH(l)?

A. -602      B. -1329      C. 1329      D. 602
- When 789 g of liquid ammonia is cooled from 82.7 °C to 25.0 °C it loses 214 kJ of heat. What is the specific heat capacity of liquid ammonia?

A. 0.213      B. 4.19      C. 0.0047      D. 2.06      E. 4.70
- Identify the correct expression that can be used to determine the change in thermal energy for the following reaction:

$$\text{C}_2\text{H}_2(\text{g}) + \frac{13}{2}\text{O}_2(\text{g}) \rightarrow 4\text{CO}_2(\text{g}) + 5\text{H}_2\text{O}(\text{l}) + \text{heat?}$$

A.  $\Delta H^\circ = [(4 \text{ mol})(\Delta H_f^\circ \text{CO}_2(\text{g})) + (5 \text{ mol})(\Delta H_f^\circ \text{H}_2\text{O}(\text{l}))] + [(1 \text{ mol})(\Delta H_f^\circ \text{C}_2\text{H}_2(\text{g})) + (\frac{13}{2} \text{ mol})(\Delta H_f^\circ \text{O}_2(\text{g}))]$       B.  $\Delta H^\circ = [(1 \text{ mol})(\Delta H_f^\circ \text{C}_2\text{H}_2(\text{g})) + (\frac{13}{2} \text{ mol})(\Delta H_f^\circ \text{O}_2(\text{g}))] - [(4 \text{ mol})(\Delta H_f^\circ \text{CO}_2(\text{g})) + (5 \text{ mol})(\Delta H_f^\circ \text{H}_2\text{O}(\text{l}))]$

C.  $\Delta H^\circ = [(1 \text{ mol})(\Delta H_f^\circ \text{C}_2\text{H}_2(\text{g})) + (\frac{13}{2} \text{ mol})(\Delta H_f^\circ \text{O}_2(\text{g}))] - [(4 \text{ mol})(\Delta H_f^\circ \text{CO}_2(\text{g})) + (5 \text{ mol})(\Delta H_f^\circ \text{H}_2\text{O}(\text{l}))]$

D.  $\Delta H^\circ = [(4 \text{ mol})(\Delta H_f^\circ \text{CO}_2(\text{g})) + (5 \text{ mol})(\Delta H_f^\circ \text{H}_2\text{O}(\text{l}))] - [(1 \text{ mol})(\Delta H_f^\circ \text{C}_2\text{H}_2(\text{g})) + (\frac{13}{2} \text{ mol})(\Delta H_f^\circ \text{O}_2(\text{g}))]$

E.  $\Delta H^\circ = [(1 \text{ mol})(\Delta H_f^\circ \text{CO}_2(\text{g})) + (\frac{4}{5} \text{ mol})(\Delta H_f^\circ \text{H}_2\text{O}(\text{l}))] - [(1 \text{ mol})(\Delta H_f^\circ \text{C}_2\text{H}_2(\text{g})) + (\frac{13}{2} \text{ mol})(\Delta H_f^\circ \text{O}_2(\text{g}))]$
- For which of the following statements is correct for the reaction

$$2\text{H}_2\text{O}_2(\text{g}) \rightarrow \text{H}_2(\text{g}) + 2\text{O}_2(\text{g}) + 66.4 \text{ kJ}$$

A. exothermic, potential energy greater in reactants  
 B. exothermic, potential energy greater in products  
 C. endothermic, potential energy greater in products  
 D. endothermic, potential energy less in reactants  
 E. endothermic, potential energy less in products

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**K Morrison**



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web by the end of this section you will be able to do the following describe the behavior of chromosomes during meiosis and the differences between the first and second meiotic divisions describe the cellular events that take place during meiosis explain the differences between meiosis and mitosis

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[the mark of vishnu and other stories by khushwant singh](#) - Feb 26 2022

web nov 25 2020 the mark of vishnu and other stories by khushwant singh 1950 saturn press edition microform in english

[the mark of vishnu summary studienet dk](#) - Jan 28 2022

web summary the short story the mark of vishnu by khushwant singh begins with gunga ram a servant leaving a saucer of milk for a cobra he calls the kala nag outside a hole in the wall he tells the schoolboys the milk is gone by morning and this offering protects the household from the snake

**the mark of vishnu characters studienet dk** - Dec 27 2021

web the main characters in the short story the mark of vishnu by khushwant singh are gunga ram and the four schoolboys gunga ram is a servant in the household where the schoolboys live the four brothers go to school and receive a formal education

[the mark of vishnu by khushwant singh the sitting bee](#) - May 12 2023

web in the mark of vishnu by khushwant singh we have the theme of tradition modernity faith loyalty respect and change taken from his collected stories collection the story is narrated in the first person by a young unnamed boy and after reading the story the reader realises that singh may be exploring the theme of tradition

**the mark of vishnu themes and message prime study guides** - Apr 30 2022

web the theme of the short story the mark of vishnu by khushwant singh is superstition and ignorance the main character

gunga ram is both superstitious and ignorant he believes that by leaving a saucer full of milk for the snake to drink he is protecting the household from snake bites

**the mark of vishnu characters prime study guides** - Mar 30 2022

web the main characters in the short story the mark of vishnu by khushwant singh are gunga ram and the four schoolboys gunga ram is a servant in the household where the schoolboys live the four brothers go to school and receive a formal education since gunga ram is a servant the schoolboys make fun of him for his lack of education and