

Exam 1 – version 1

1. How is the substance carbon monoxide classified? (Hint: It has two elements that have a fixed relative composition. This substance cannot be broken down by physical means.)
- a) Element **b) Compound** c) Heterogeneous mixture d) Homogeneous mixture
5. A men's regulation basketball has a volume of 43.4 m^3 . What is this volume in cubic centimeters, cm^3 ?
- a) $7.33 \times 10^7 \text{ cm}^3$ b) $1.10 \times 10^7 \text{ cm}^3$ c) $2.65 \times 10^7 \text{ cm}^3$
d) $8.71 \times 10^7 \text{ cm}^3$ e) $1.34 \times 10^7 \text{ cm}^3$
9. A scientist measured the density of water three times and compared these results with the accepted value which is 1.000 g/mL . The following results were obtained:

Experiment #1: 0.904 g/mL
Experiment #2: 0.902 g/mL
Experiment #3: 0.906 g/mL

How would these results be characterized?

- a) high accuracy and high precision
b) high accuracy and low precision
c) low accuracy and high precision
d) low accuracy and low precision

How to convert from °C to F and vice versa

Review names of elements

Locate metalloids–metals etc. on periodic table

18. Which of the following elements would be expected to have the most similar chemical and physical properties?
- a) **Magnesium and Calcium** b) Silver and Iron c) Oxygen and Nitrogen
d) Phosphorous and Boron e) Neon and Fluorine

Review A –mass#, Z –atomic #, q –charge, electrons

23. What is the systematic name for the compound, ZnCl_2 ?
- a) Zinc Chloride b) Zinc(II) chloride c) Zinc dichloride
d) Zinc chloride e) Zinc Chlorine

names of polyatomic ions, naming inorganic compounds

write the metric prefixes

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