

CHEMISTRY CHEAT SHEET

CHEMICAL REACTIONS

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Anatomy of a Chemical Reaction Equation



Type of Reaction	General Format
Combustion	$C_xH_y + O_2 \rightarrow CO_2 + H_2O$
Decomposition	$AB \rightarrow A + B$
Double Displacement	$AB + CD \rightarrow AD + CB$
Single Displacement	$A + BC \rightarrow AC + B$
Redox	$A + B \rightarrow A^{+} + B^{-}$
Acid-Base	$HA + B \rightarrow A^{-} + HB^{+}$
Neutralization	$HA + BOH \rightarrow AB + H_2O$

- #### Indicators of a Chemical Reaction
- Color change
 - Temperature change
 - Gas production
 - Formation of a precipitate
 - Change in pH
 - Change in mass

Exothermic and Endothermic Reactions

Exothermic Reaction - releases heat and light. $\Delta H < 0$

$$A + B \rightarrow C + D + \text{Heat}$$

Endothermic Reaction - absorbs heat. $\Delta H > 0$

$$A + B + \text{Heat} \rightarrow C + D$$

Common Reaction Arrow Symbols

$\xrightarrow{\Delta}$	Heat (exothermic)
$\xrightarrow{h\nu}$	Light (photochemical)
\xrightarrow{Pt}	Pt catalyst
\xrightarrow{Ni}	Ni catalyst
$\xrightarrow{H^+}$	Acid catalyst
$\xrightarrow{OH^-}$	Base catalyst
$\xrightarrow{[O]}$	Oxidation (loss of electrons)
$\xrightarrow{[R]}$	Reduction (gain of electrons)

Graphical Representations of Exothermic and Endothermic Reactions



Chemical Reactions Cheat Sheet

Jessica J Manson



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