



Name: _____ Date: _____ Class period: _____

Unit: Chemical Quantities "Test Review" - Wksh #8 ANSWER KEY



What to study?

- Homework: Worksheets #1-#7
- Notes: Mole Calculations → Determining Molecular Formulas
- Bell Ringers: All bell ringers in this unit
- All quizzes

What will the questions be like? What can I expect?

This test will be just like the previous quizzes you have taken.

- Calculations

Sample questions/Important Topics

Directions: Use dimensional analysis to perform the following calculations. Show all of your work, include your units and round answers to the correct number of significant figures.

1. How many atoms are equal to 4.61 moles of carbon? **2.78×10^{24} atoms C**

2. How many moles are equal to 1.27×10^{23} molecules of H_2O ? **0.211 mol H_2O**

3. How many moles are equal to 47.2 g of K_2O ? **0.501 mol K_2O**

4. What mass of $Ca(NO_3)_2$ is equal to 0.75 moles of this substance? **120g $Ca(NO_3)_2$**

Chemical Quantities Guided Practice Answers

Iris Stovall, Steven S. Zumdahl



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