

কুমিল্লা বোর্ড

১২। বিপুল পরিমাণে কয়েক ভেঁটি HCl দ্রবণ করলে K_w এর মান হবে কত?

- (ক) 1×10^{-16}
 (খ) 1×10^{-14}
 (গ) 1×10^{-13}
 (ঘ) 1×10^{-7}

১৩। কেন্দ্রীয় পরমাণুর অষ্টক পূর্ণ হচ্ছে—

- i. H_2O
 ii. BCl_3
 iii. NCl_3

নিচের কোনটি সঠিক?

- (ক) i ও ii (খ) ii ও iii
 (গ) i ও iii (ঘ) i, ii ও iii

১৪। মানুষের রক্তের pH মান কত?

- (ক) 7-4
 (খ) 7-0
 (গ) 5-5
 (ঘ) 3-2

১৫। আয়নীকরণ বিভবের সঠিক ক্রম কোনটি?

- (ক) $N > O > F$
 (খ) $Br < Cl < F$
 (গ) $K < N < Li$
 (ঘ) $Be > Mg > C$

১৬। PH_4Cl বোঁতে বিদ্যমান বন্ধন—

- i. আয়নিক
 ii. সমযোজী
 iii. সন্ধিবেশ

নিচের কোনটি সঠিক?

- (ক) i ও ii (খ) i ও iii
 (গ) ii ও iii (ঘ) i, ii ও iii

১৭। নিচের কোন বিক্রয়টি তিনপদার্থ?

- (ক) $HNO_3 + CH_3COOH$
 (খ) $CH_3COOH + H_2O$
 (গ) $HCl + CH_3COOH$
 (ঘ) $H_2SO_4 + CH_3COOH$

১৮। কোনটিতে পোল্যারায়ন বেশি ঘটিবে?

- (ক) $BeCl_2$
 (খ) $MgCl_2$
 (গ) $LiCl$
 (ঘ) $NaCl$

১৯। বরফের একটি অণুরেণ পরমাণুতে কতটি H-বন্ধন বিদ্যমান?

- (ক) 2
 (খ) 3
 (গ) 4
 (ঘ) 5

২০। $_{28}Ni$ এর কক্ষকগুলো ইলেকট্রনের ক্ষেত্রে $(n + l) = 4$ হলে (এখানে, $l = 1$)?

- (ক) 4
 (খ) 5
 (গ) 6
 (ঘ) 7

২১। কোনটি দুশ্চয়্যে বর্ণিত?

- (ক) লাইম্যান সিরিজ
 (খ) বামার সিরিজ
 (গ) প্যাঞ্চেন সিরিজ
 (ঘ) ব্র্যাকট সিরিজ

২২। জাল প্লাসপোর্ট শনাক্তকরণে কোনটি ব্যবহার করা হয়?

- (ক) γ -রশ্মি (খ) X-রশ্মি
 (গ) IR-রশ্মি (ঘ) UV-রশ্মি

নিচের উদ্বীপকটি পড় এবং ২৩ ও ২৪ নং প্রশ্নের উত্তর দাও :

বোঁল	ইলেকট্রন বিন্যাস	অঙ্কিত কণ্যাত্বকতা
L	ns^2	—
M	$(n+1)s^2$	—
N	ns^2sp^5	3-0

এখানে, $n = 3$

২৩। N_2 অণুর ক্ষেত্রে কোনটি প্রযোজ্য?

- (ক) দুক জোড় ইলেকট্রন একটি
 (খ) অপোলার সমযোজী
 (গ) পোলার সমযোজী
 (ঘ) sp সংকরণ বিশিষ্ট

২৪। LN_2 এবং MN_2 এর মধ্যে MN_2 অধিক—

- i. গলনাঙ্কবিশিষ্ট
 ii. পানিতে দ্রবণীয়
 iii. সমযোজী

নিচের কোনটি সঠিক?

- (ক) i ও ii (খ) ii ও iii
 (গ) i ও iii (ঘ) i, ii ও iii

২৫। সবচেয়ে দুর্বল এসিড কোনটি?

- (ক) H_3BO_3 (খ) HNO_3
 (গ) H_3PO_3 (ঘ) $HClO_3$

Ref: হাজারী সার
 Edition-2023; Page-637
 Ka এর মান $[5.8 \times 10^{-7}]$
 $[1 \times 10^{-7}]$

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Alan William Mackaill



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 - The Rise of Digital Reading Chemistry 1st Paper Question Hsc 2013 Bing
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry 1st Paper Question Hsc 2013 Bing
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
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 - Popular eBook Platforms
 - Features to Look for in an Chemistry 1st Paper Question Hsc 2013 Bing
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry 1st Paper Question Hsc 2013 Bing
 - Personalized Recommendations

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 - Chemistry 1st Paper Question Hsc 2013 Bing Budget-Friendly Options
- 6. Navigating Chemistry 1st Paper Question Hsc 2013 Bing eBook Formats
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 - Chemistry 1st Paper Question Hsc 2013 Bing Compatibility with Devices
 - Chemistry 1st Paper Question Hsc 2013 Bing Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry 1st Paper Question Hsc 2013 Bing
 - Highlighting and Note-Taking Chemistry 1st Paper Question Hsc 2013 Bing
 - Interactive Elements Chemistry 1st Paper Question Hsc 2013 Bing
- 8. Staying Engaged with Chemistry 1st Paper Question Hsc 2013 Bing
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry 1st Paper Question Hsc 2013 Bing
- 9. Balancing eBooks and Physical Books Chemistry 1st Paper Question Hsc 2013 Bing
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry 1st Paper Question Hsc 2013 Bing
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry 1st Paper Question Hsc 2013 Bing
 - Setting Reading Goals Chemistry 1st Paper Question Hsc 2013 Bing
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry 1st Paper Question Hsc 2013 Bing
 - Fact-Checking eBook Content of Chemistry 1st Paper Question Hsc 2013 Bing

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 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
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