



Lab 15a: Investigating Chemical Equilibrium

Use the following information to answer questions 1-10. The following information is provided for your reference. The following information is provided for your reference.

The reaction between Fe^{3+} and SCN^{-} is reversible. The reaction is shown below.

The reaction between Fe^{3+} and SCN^{-} is reversible. The reaction is shown below.

The reaction between Fe^{3+} and SCN^{-} is reversible. The reaction is shown below.

1. Write the balanced chemical equation for the reaction between Fe^{3+} and SCN^{-} .

2. Write the equilibrium constant expression for the reaction between Fe^{3+} and SCN^{-} .

3. Write the equilibrium constant expression for the reaction between Fe^{3+} and SCN^{-} .

4. Write the equilibrium constant expression for the reaction between Fe^{3+} and SCN^{-} .

5. Write the equilibrium constant expression for the reaction between Fe^{3+} and SCN^{-} .

6. Write the equilibrium constant expression for the reaction between Fe^{3+} and SCN^{-} .

7. Write the equilibrium constant expression for the reaction between Fe^{3+} and SCN^{-} .

8. Write the equilibrium constant expression for the reaction between Fe^{3+} and SCN^{-} .

9. Write the equilibrium constant expression for the reaction between Fe^{3+} and SCN^{-} .

10. Write the equilibrium constant expression for the reaction between Fe^{3+} and SCN^{-} .

11. Write the equilibrium constant expression for the reaction between Fe^{3+} and SCN^{-} .

12. Write the equilibrium constant expression for the reaction between Fe^{3+} and SCN^{-} .

13. Write the equilibrium constant expression for the reaction between Fe^{3+} and SCN^{-} .

14. Write the equilibrium constant expression for the reaction between Fe^{3+} and SCN^{-} .

15. Write the equilibrium constant expression for the reaction between Fe^{3+} and SCN^{-} .

16. Write the equilibrium constant expression for the reaction between Fe^{3+} and SCN^{-} .

17. Write the equilibrium constant expression for the reaction between Fe^{3+} and SCN^{-} .

18. Write the equilibrium constant expression for the reaction between Fe^{3+} and SCN^{-} .

19. Write the equilibrium constant expression for the reaction between Fe^{3+} and SCN^{-} .

20. Write the equilibrium constant expression for the reaction between Fe^{3+} and SCN^{-} .

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