

COMMON TYPES OF CHEMICAL REACTIONS

SYNTHESIS

Multiple reactants, which can be simple elements or compounds, combine together to form a single compound.



DECOMPOSITION

A compound breaks down into two or more simpler substances. Decomposition reactions are classified into thermal, electrolytic, and photo.



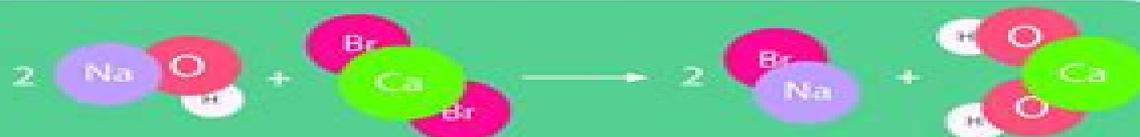
SINGLE-REPLACEMENT

One element is substituted for another element in a compound, generating a new compound and a pure element.



DOUBLE-REPLACEMENT

Two ionic compounds exchange cations or anions to form two new compounds. Forming a precipitate can help drive the reaction to the right.



COMBUSTION

A fuel, typically a hydrocarbon, reacts with oxygen gas to form carbon dioxide and water, which generates heat and light.



NEUTRALIZATION

Acid and base combine to neutralize each other in aqueous solutions, usually resulting in a salt and water.



Chemical Reactions Chem Think Answers

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Efremovich Zaikov**

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In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Chemical Reactions Chem Think Answers**," an enthralling opus penned by a very acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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