

## Types of Chemical Reactions Lab Answer Key

### Chemical Reaction #1

<i>Reactants (write in words)</i>	hydrochloric acid and magnesium
<i>Observations</i>	accept most responses
<i>Type of Reaction</i>	single replacement
<i>Predict Products (write in words)</i>	magnesium chloride and hydrogen gas
<i>Equation (write in symbols)</i>	$2\text{HCl}_{(aq)} + \text{Mg}_{(s)} \rightarrow \text{MgCl}_{2(aq)} + \text{H}_{2(g)}$

### Chemical Reaction #2

<i>Reactants (write in words)</i>	copper metal and oxygen gas
<i>Observations</i>	accept most responses
<i>Type of Reaction</i>	synthesis
<i>Predict Products (write in words)</i>	copper oxide
<i>Equation (write in symbols)</i>	$2\text{Cu}_{(s)} + \text{O}_{2(g)} \rightarrow 2\text{CuO}_{(s)}$

### Chemical Reaction #3

<i>Reactants (write in words)</i>	methane gas and oxygen gas
<i>Observations</i>	accept most responses
<i>Type of Reaction</i>	combustion
<i>Predict Products (write in words)</i>	carbon dioxide and water
<i>Equation (write in symbols)</i>	$\text{CH}_{4(g)} + 2\text{O}_{2(g)} \rightarrow \text{CO}_{2(g)} + 2\text{H}_2\text{O}_{(g)}$

### Chemical Reaction #4

<i>Reactants (write in words)</i>	hydrogen peroxide
<i>Observations</i>	accept most responses
<i>Type of Reaction</i>	decomposition
<i>Predict Products (write in words)</i>	water and oxygen gas
<i>Equation (write in symbols)</i>	$2\text{H}_2\text{O}_{2(l)} \rightarrow 2\text{H}_2\text{O}_{(l)} + \text{O}_{2(g)}$

# Chemical Reaction Lab Answer Key Chemfiesta

**Darrell D. Ebbing**



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**Passing the North Carolina End of Course Test for Chemistry** Liz Thompson, 2007

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