

Chemistry Calculations - Study-Guide

1. How many molecules are there in 2.00 grams of CO_2 ?
 B) 1.20×10^{23} molecules
2. How many formula units are there in 1.0 mole of CaCl_2 ?
 B) 6.02×10^{23}
3. How many molecules are there in 8 molecules of hydrogen peroxide, H_2O_2 (g)?
 B) 8 molecules
4. The gram molecular mass of oxygen is 32. B) 32 g/mol
5. The gram atomic mass of oxygen is 16. B) 16 g/mol
6. What is the gram formula mass of CaCl_2 ? B) 110.98 g/mol
7. What is the number of moles in 3.0 grams of NaCl ?
 B) 0.0518 moles NaCl
8. The gram formula mass of H_2O is 18 g. What is the number of moles in 36 g of H_2O ?
 C) 2.00 moles H_2O ; 1.00 moles H_2O ; 0.50 moles H_2O ; 0.25 moles H_2O
 B) 2.00 mol
9. A mole is a measure of molecular composition with 6.023 moles of H_2O there is approximately what is there for composition? B) 6.023×10^{23}
10. What is the percent composition of aluminum in Al_2O_3 ?
 B) 52.9% Al
11. What is the empirical formula of a compound that is 80% carbon and 20% hydrogen by weight?
 B) C_4H_{10}

Chemical Quantities Study Guide Answers

S Ben Porath



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