

Balancing Equations Race

- 1) $\underline{\quad}$ $\text{C}_3\text{H}_8 + \underline{5} \text{O}_2 \rightarrow \underline{3} \text{CO}_2 + \underline{4} \text{H}_2\text{O}$
- 2) $\underline{\quad}$ $\text{Al} + \underline{\quad} \text{Fe}_3\text{N}_2 \rightarrow \underline{\quad} \text{AlN} + \underline{\quad} \text{Fe}$
- 3) $\underline{\quad}$ $\text{Na} + \underline{\quad} \text{Cl}_2 \rightarrow \underline{\quad} \text{NaCl}$
- 4) $\underline{\quad}$ $\text{H}_2\text{O}_2 \rightarrow \underline{\quad} \text{H}_2\text{O} + \underline{\quad} \text{O}_2$
- 5) $\underline{\quad}$ $\text{C}_6\text{H}_{12}\text{O}_6 + \underline{\quad} \text{O}_2 \rightarrow \underline{\quad} \text{H}_2\text{O} + \underline{\quad} \text{CO}_2$
- 6) $\underline{\quad}$ $\text{H}_2\text{O} + \underline{\quad} \text{CO}_2 \rightarrow \underline{\quad} \text{C}_2\text{H}_6 + \underline{\quad} \text{O}_2$
- 7) $\underline{\quad}$ $\text{NaClO}_3 \rightarrow \underline{\quad} \text{NaCl} + \underline{\quad} \text{O}_2$
- 8) $\underline{4} (\text{NH}_4)_3\text{PO}_4 + \underline{3} \text{Pb}(\text{NO}_3)_4 \rightarrow \underline{\quad} \text{Pb}_3(\text{PO}_4)_4 + \underline{12} \text{NH}_4\text{NO}_3$
- 9) $\underline{\quad} \text{BF}_3 + \underline{\quad} \text{Li}_2\text{SO}_3 \rightarrow \underline{\quad} \text{B}_2(\text{SO}_3)_3 + \underline{\quad} \text{LiF}$
- 10) $\underline{\quad} \text{C}_2\text{H}_6 + \underline{\quad} \text{O}_2 \rightarrow \underline{\quad} \text{CO}_2 + \underline{\quad} \text{H}_2\text{O}$
- 11) $\underline{\quad} \text{CaCO}_3 + \underline{\quad} \text{H}_3\text{PO}_4 \rightarrow \underline{\quad} \text{Ca}_3(\text{PO}_4)_2 + \underline{\quad} \text{H}_2\text{CO}_3$
- 12) $\underline{\quad} \text{Ag}_2\text{S} \rightarrow \underline{\quad} \text{Ag} + \underline{\quad} \text{S}_8$
- 13) $\underline{\quad} \text{KBr} + \underline{\quad} \text{Fe}(\text{OH})_3 \rightarrow \underline{\quad} \text{KOH} + \underline{\quad} \text{FeBr}_3$
- 14) $\underline{\quad} \text{KNO}_3 + \underline{\quad} \text{H}_2\text{CO}_3 \rightarrow \underline{\quad} \text{K}_2\text{CO}_3 + \underline{\quad} \text{HNO}_3$
- 15) $\underline{\quad} \text{Pb}(\text{OH})_4 + \underline{\quad} \text{Cu}_2\text{O} \rightarrow \underline{\quad} \text{PbO}_2 + \underline{\quad} \text{CuOH}$
- 16) $\underline{\quad} \text{Cr}(\text{NO}_2)_2 + \underline{\quad} (\text{NH}_4)_2\text{SO}_4 \rightarrow \underline{\quad} \text{CrSO}_4 + \underline{\quad} \text{NH}_4\text{NO}_2$
- 17) $\underline{\quad} \text{KOH} + \underline{\quad} \text{Co}_3(\text{PO}_4)_2 \rightarrow \underline{\quad} \text{K}_3\text{PO}_4 + \underline{\quad} \text{Co}(\text{OH})_2$
- 18) $\underline{\quad} \text{Sn}(\text{NO}_2)_4 + \underline{\quad} \text{Pt}_3\text{N}_4 \rightarrow \underline{\quad} \text{Sn}_3\text{N}_4 + \underline{\quad} \text{Pt}(\text{NO}_2)_4$
- 19) $\underline{\quad} \text{B}_2\text{Br}_8 + \underline{\quad} \text{HNO}_3 \rightarrow \underline{\quad} \text{B}(\text{NO}_3)_3 + \underline{\quad} \text{HBr}$
- 20) $\underline{\quad} \text{ZnS} + \underline{\quad} \text{AlP} \rightarrow \underline{\quad} \text{Zn}_3\text{P}_2 + \underline{\quad} \text{Al}_2\text{S}_3$

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