

# ChemActivity 1

To the Student

## The Nuclear Atom

(What Is an Atom?)

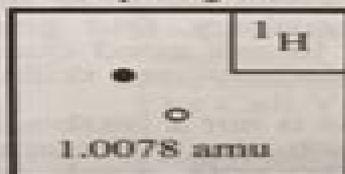
**Model: Schematic Diagrams for Various Atoms and Ions.**

- electron (-)
- proton (+)
- neutron (no charge)

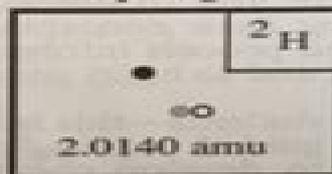
$$1 \text{ amu} = 1.6606 \times 10^{-24} \text{ g}$$

The **nucleus** of an atom contains the protons and the neutrons.

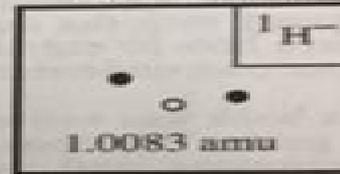
Hydrogen



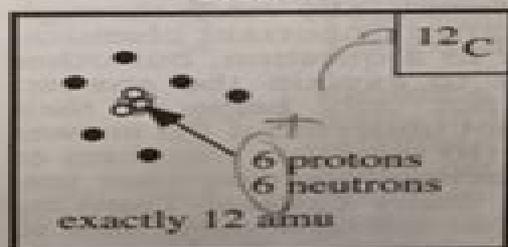
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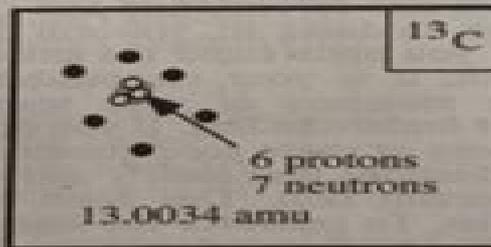
Hydrogen ion



Carbon



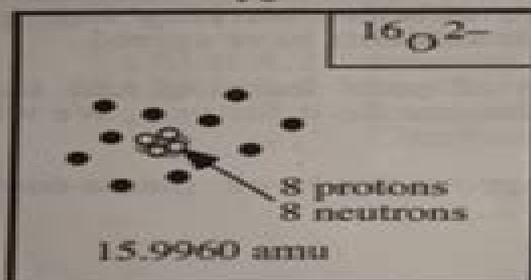
Carbon



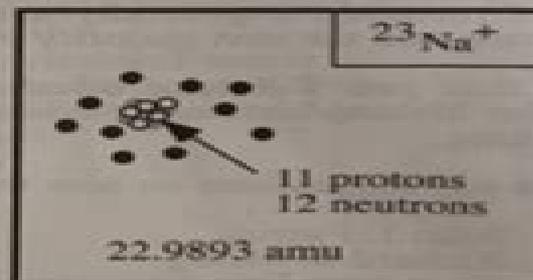
Carbon ion



Oxygen ion



Sodium ion



${}^1\text{H}$  and  ${}^2\text{H}$  are isotopes of hydrogen.

${}^{12}\text{C}$  and  ${}^{13}\text{C}$  are isotopes of carbon.

# Chemactivity The Nuclear Atom Key

**M Tight**



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