

Multiple Choice Questions for AQA AS Chemistry

FREE WORKSHEETS

AQA A Level Chemistry Physical Chemistry 1

cheMCQ

Amount of Substance

This test consists of TEN multiple choice questions. You should choose ONE answer for each question.

- How many molecules are present in 15.0 g of PH_3 ? (Avogadro's number is 6.022×10^{23})
 - A 1.2×10^{24}
 - B 1.355×10^{24}
 - C 1.355×10^{23}
 - D 1.37×10^{24}
- What is the mass number of oxygen?
 - A 16
 - B 15
 - C 17
 - D 8
- The density of pure water is 1000 g dm^{-3} . What is the concentration of pure water in mol dm^{-3} ?
 - A 1 mol dm^{-3}
 - B 55 mol dm^{-3}
 - C 32 mol dm^{-3}
 - D 18 mol dm^{-3}
- Calculate the mass of KOH that should be dissolved in 500 cm^3 water to give a solution with a concentration of 0.6 mol dm^{-3}
 - A 0.0673 g
 - B 187 g
 - C 3 g
 - D 9 g

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Multiple Choice Questions for AS and A Level Chemistry

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- A 1 dm^3 sample of Noble gas X contains 0.04471 moles. The density of the gas is 1.784 g dm^{-3} . Which gas does the sample contain?
 - A Helium
 - B Neon
 - C Argon
 - D Krypton
- Calculate the concentration of the solution that is formed by adding 50 cm^3 of water to 25 cm^3 of $0.268 \text{ mol dm}^{-3}$ HCl
 - A $0.134 \text{ mol dm}^{-3}$
 - B $0.0893 \text{ mol dm}^{-3}$
 - C 1.34 mol dm^{-3}
 - D $3.35 \times 10^{-2} \text{ mol dm}^{-3}$
- 0.05 moles of an iron salt FeCl_x reacts with AgNO_3 to form 21.5 g of AgCl . What is the value of x ?
 - A 1
 - B 2
 - C 3
 - D 4
- A The uncertainty of a balance is $\pm 0.1 \text{ g}$. What is the % uncertainty if this balance is used to measure 26.7 g of CuSO_4 ?
 - A 0.3%
 - B 0.4%
 - C 0.04%
 - D 0.03%

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Multiple Choice Questions for AS and A Level Chemistry

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- What is the formula of Aluminium Carbonate?
 - A $\text{Al}_2(\text{CO}_3)_3$
 - B $\text{Al}_2(\text{CO}_3)$
 - C $\text{Al}_2(\text{CO}_3)_2$
 - D $\text{Al}_2(\text{CO}_3)_4$
- Dinitrogen Monoxide decomposes to produce Nitrogen and Oxygen according to the following equation:

$$2\text{N}_2\text{O}(\text{g}) \rightarrow 2\text{N}_2(\text{g}) + \text{O}_2(\text{g})$$
 Assuming a constant temperature, how many times greater is the volume of the products than the volume of the reactants?
 - A 1
 - B 0.5
 - C 1.5
 - D 0.7

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Multiple Choice Questions for AS and A Level Chemistry

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Chem Tutor 8 Self Check Activity Answers

**Federation of American Societies for
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Chemical Pedagogy Keith S Taber, 2024-12-20 How should chemistry be taught in schools colleges and universities
Chemical Pedagogy discusses teaching approaches and techniques the reasoning behind them and the evidence for their effectiveness The book surveys a wide range of different pedagogic strategies and tactics that have been recommended to better engage learners and provide more effective chemistry teaching These accounts are supported by an initial introduction to some key ideas and debates about pedagogy the science of teaching Chemical Pedagogy discusses how teaching innovations can be tested to inform research based practice Through this book the author explores the challenges of carrying out valid experimental studies in education and the impediments to generalising study results to diverse teaching and learning contexts As a result the author highlights both the need to read published studies critically and the value of teachers and lecturers testing out recommended innovations in their own classrooms Chemical Pedagogy introduces core principles from research into human cognition and learning to provide a theoretical perspective on how to best teach for engagement and understanding An examination of some of the more contentious debates about pedagogy leads to the advice to seek optimally guided instruction which balances the challenge offered to learners with the level of support provided This provides a framework for discussing a wide range of teaching approaches and techniques that have been recommended to those teaching chemistry across educational levels including both those intended to replace teaching from the front and others that can be built into traditional lecture courses to enhance the learning experience *Instructor*, 1975 **Index Medicus**, 2004 Vols for 1963 include as pt 2 of the Jan issue Medical subject headings **Teaching Chemistry in Higher Education** Michael Seery, Claire Mc Donnell, 2019-07-01 Teaching Chemistry in Higher Education celebrates the contributions of Professor Tina Overton to the scholarship and practice of teaching and learning in chemistry education Leading educators in United Kingdom Ireland and Australia three countries where Tina has had enormous impact and influence have contributed chapters on innovative approaches that are well established in their own practice Each chapter introduces the key education literature underpinning the approach being described Rationales are discussed in the context of attributes and learning outcomes desirable in modern chemistry curricula True to Tina's personal philosophy chapters offer pragmatic and useful guidance on the implementation of innovative teaching approaches drawing from the authors experience of their own practice and evaluations of their implementation Each chapter also offers key guidance points for implementation in readers own settings so as to maximise their adaptability Chapters are supplemented with further reading and supplementary materials on the book's website overtonfestschrift.wordpress.com Chapter topics include innovative approaches in facilitating group work problem solving context and problem based learning embedding transferable skills and laboratory education all themes relating to the scholarly interests of Professor Tina Overton About the Editors Michael Seery is Professor of Chemistry Education at the University of Edinburgh and is Editor of Chemistry Education Research and

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